

Thank you, PER Community!

Physics Pedagogy

- Peer instruction
 - Pre-reading quizzes
 - Conceptual questions with clickers during class
 - Concepts on exams
- Computer graphics demos
- Backwards course design
- Project-based labs
- Focus on individuals
 - Learn and use names
 - Encourage use of office hours
 - Informal interactions

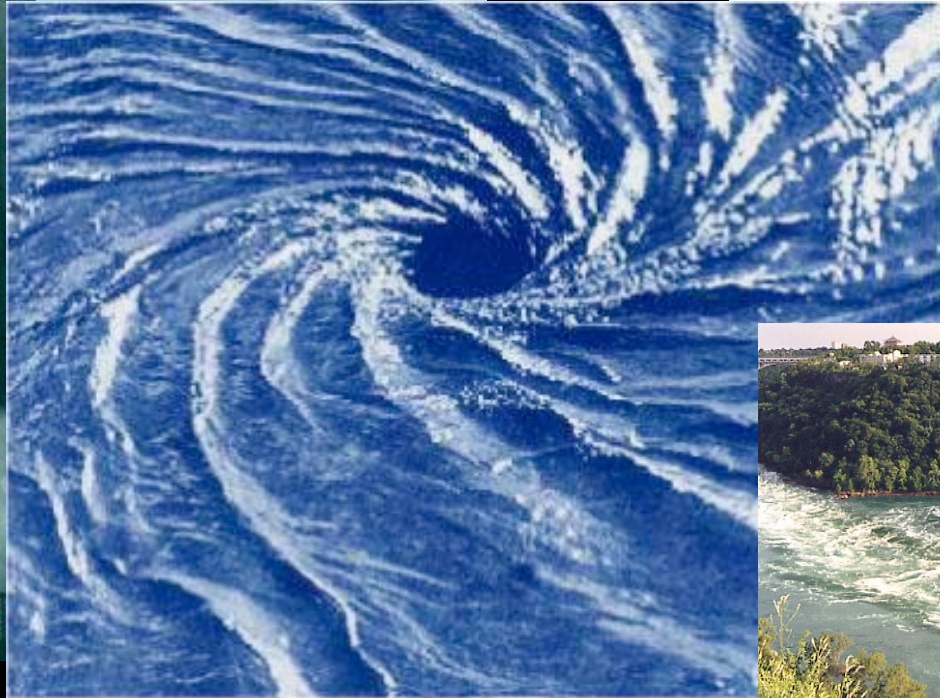
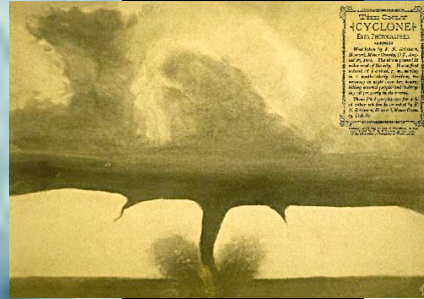
Specific Thanks to:

- Tobias Beetz
- Pat Burchat
- Marcelo Clerici-Arias
- Robyn Dunbar
- David Goldhaber-Gordon
- Chaya Nanavati
- Tom Woosnam
- All my students and TAs



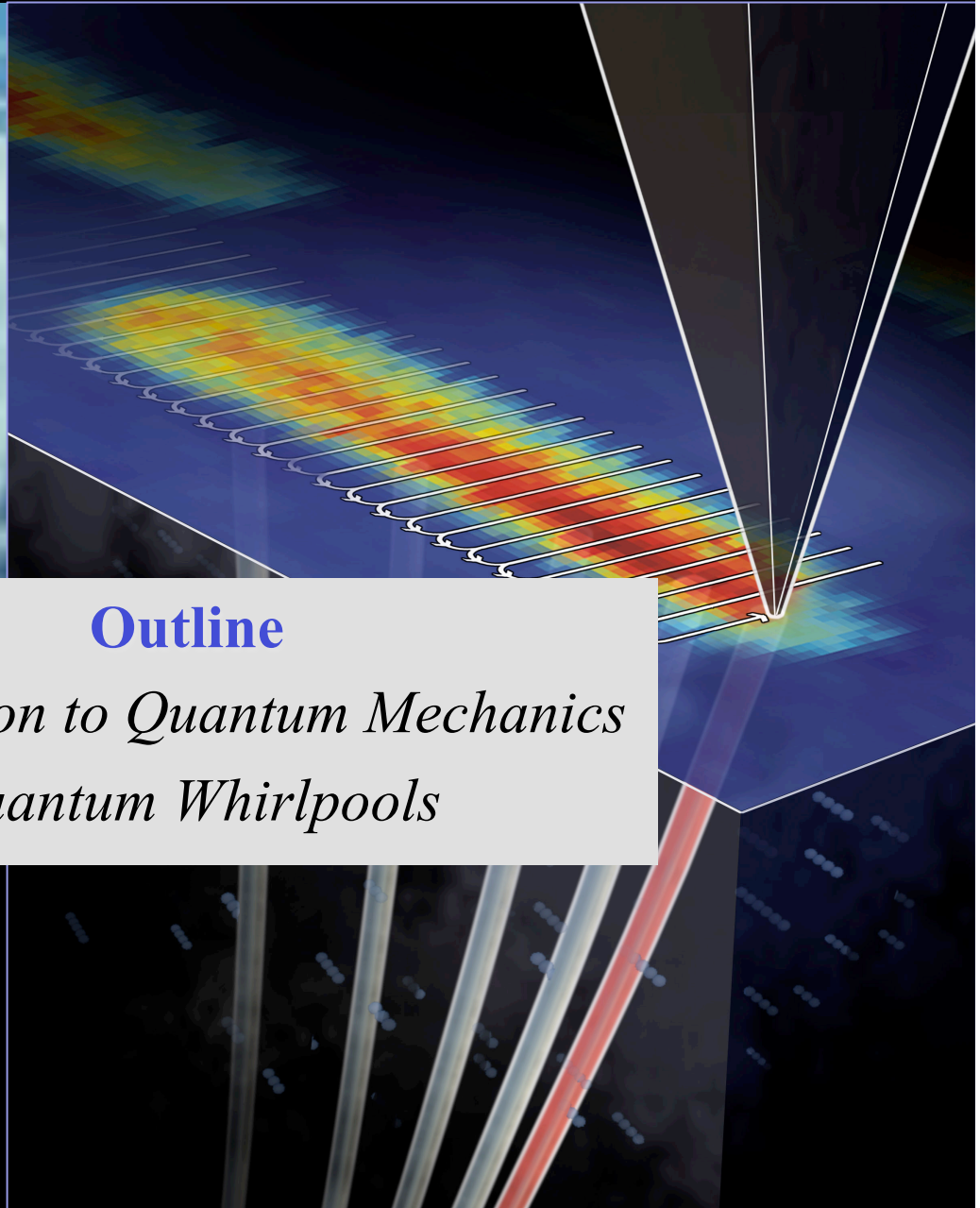
Quantum Whirlpools

Jacksonville,
1/11/10



Quantum Whirlpools

Jacksonville,
1/11/10

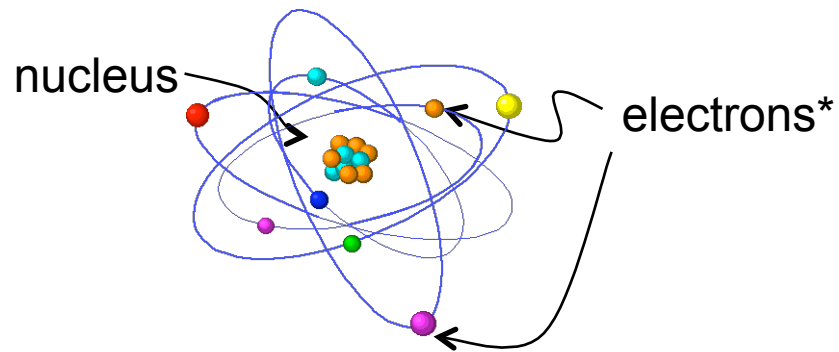


Outline

Introduction to Quantum Mechanics
Quantum Whirlpools



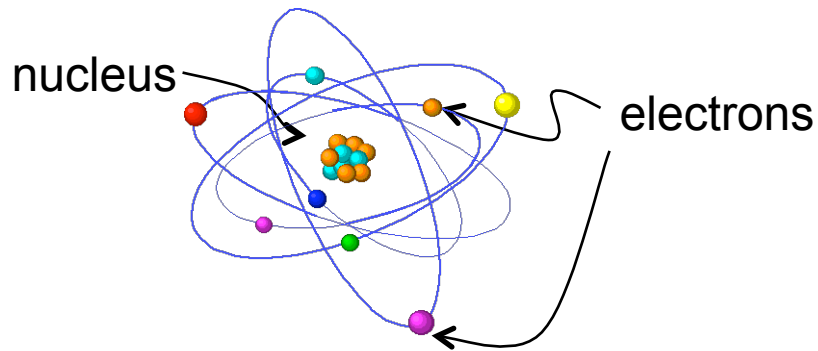
Classical view of the atom



The Economist, April 1, 2004

*An electron is a fundamental particle with mass m_e , charge e , and spin $1/2$.

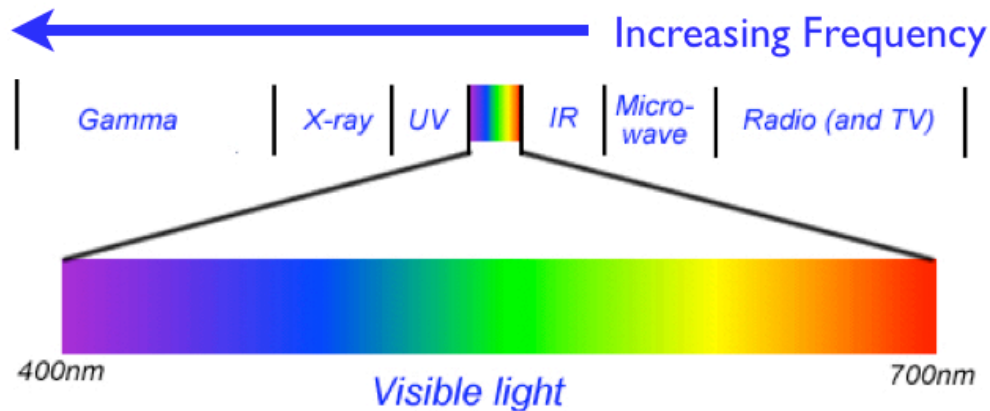
Problems with the classical view of the atom



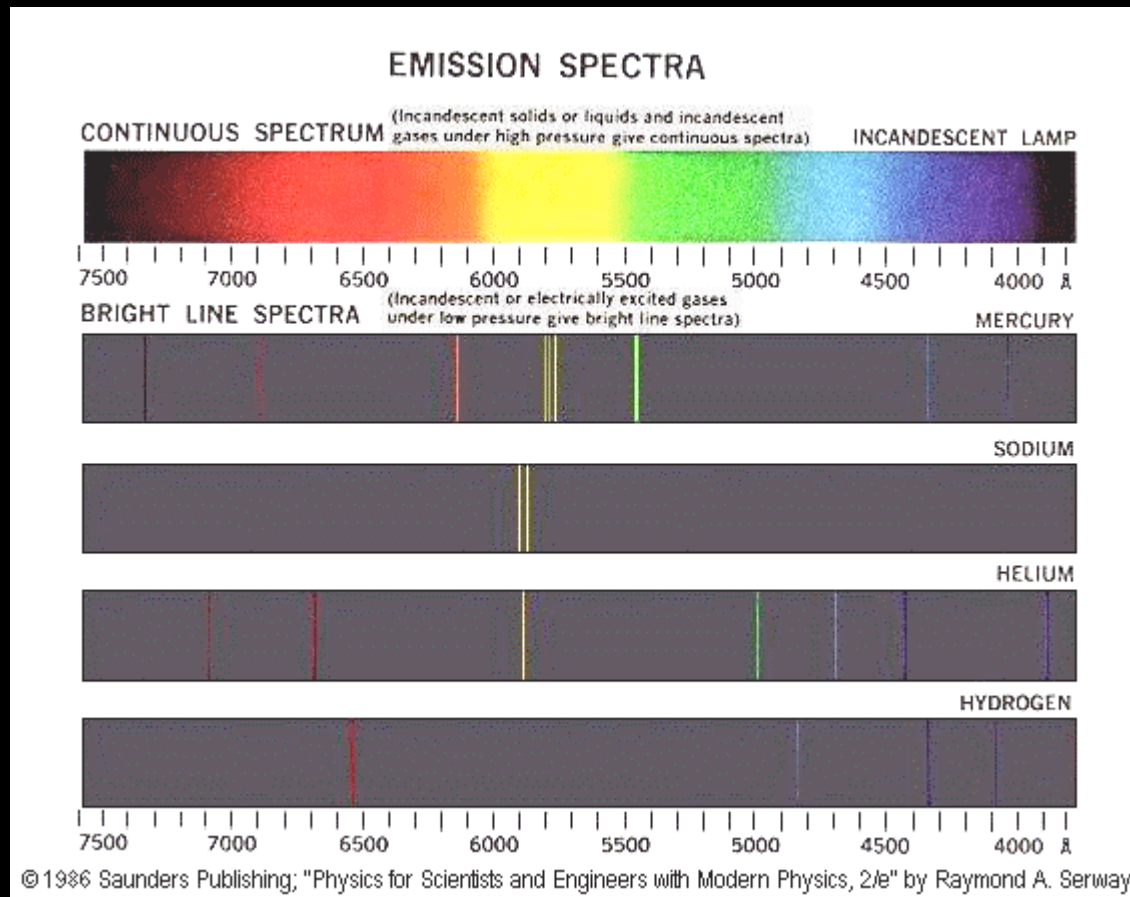
The classical (Newtonian) view of the atom would predict:

1. that the electron will spiral in and crash into the nucleus in a fraction of a second.
2. that the electron can orbit at any distance, so the energy of the atom can change by arbitrarily small amounts.

The color of light is related to its frequency:



Experimental reality: discrete (“quantized”) absorption and emission of light from atoms



continuous spectrum
of visible light

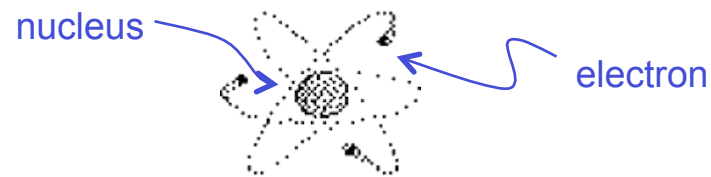
mercury (quantized)

sodium (quantized)

helium (quantized)

hydrogen (quantized)

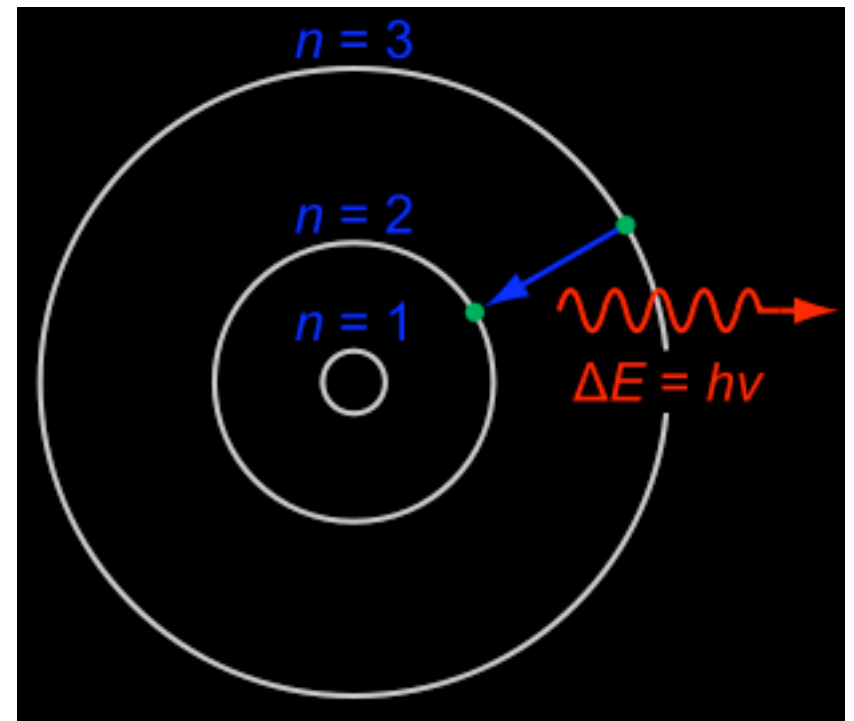
Bohr model of the hydrogen atom

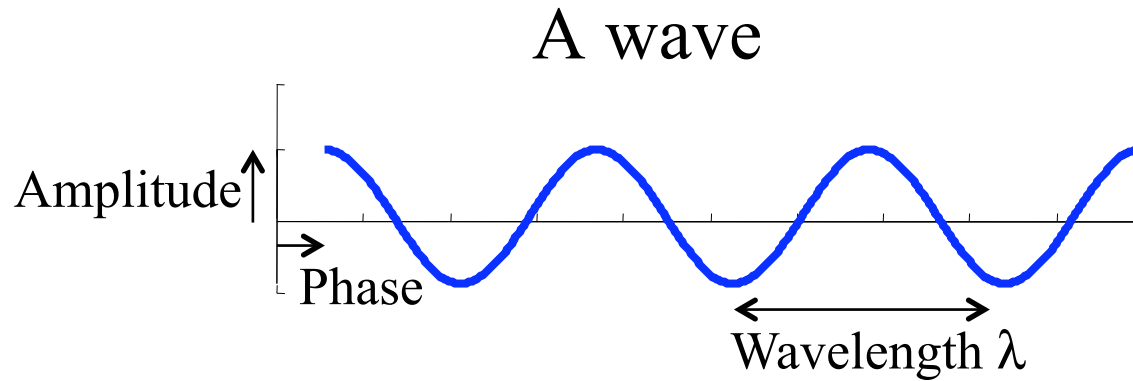


$$E_n = -\frac{13.6}{n^2} \text{ eV}$$

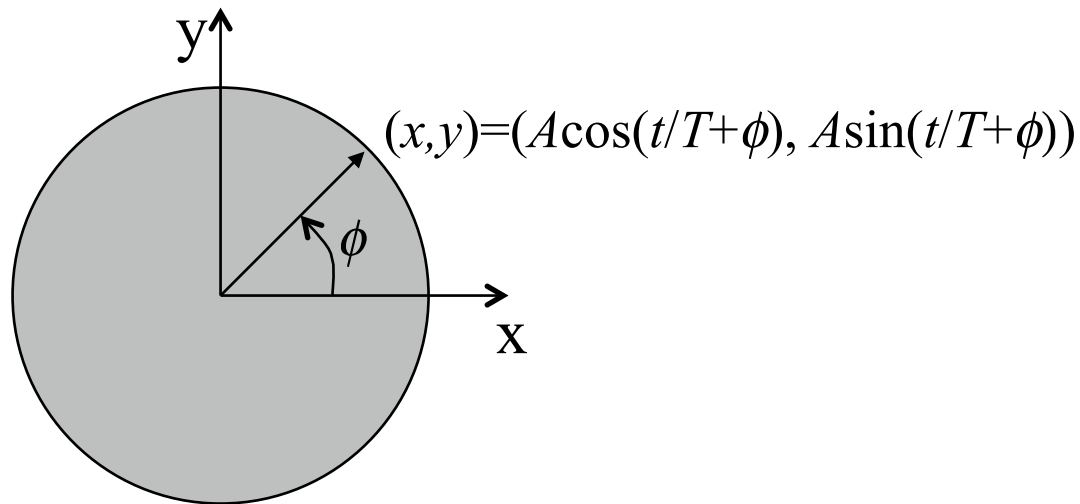
Only states with integer values of n are allowed.

The energy is *quantized*.





*Aside: the quantum mechanical wave function has an **Internal** phase:
think of a hand on a clock, or an oven timer*



The wave function

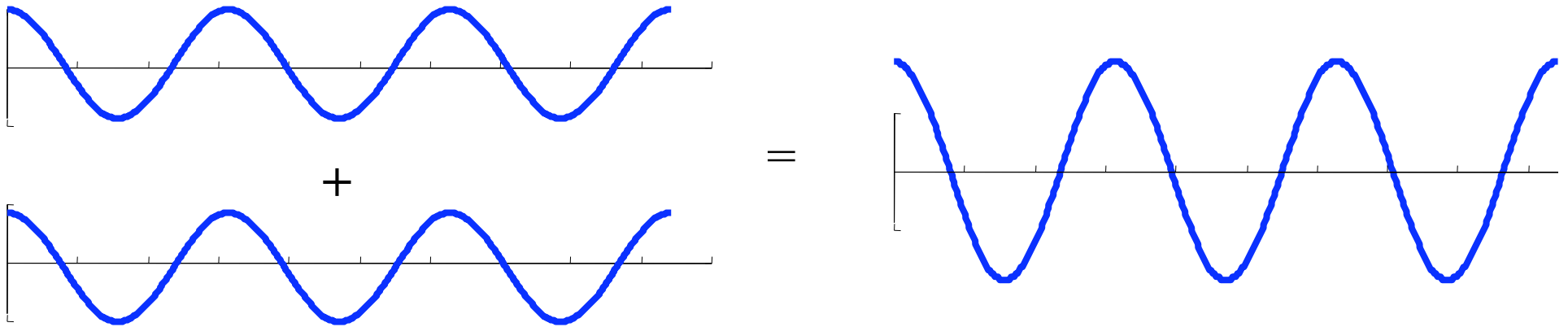
$$\Psi(r,t)$$

has an internal phase

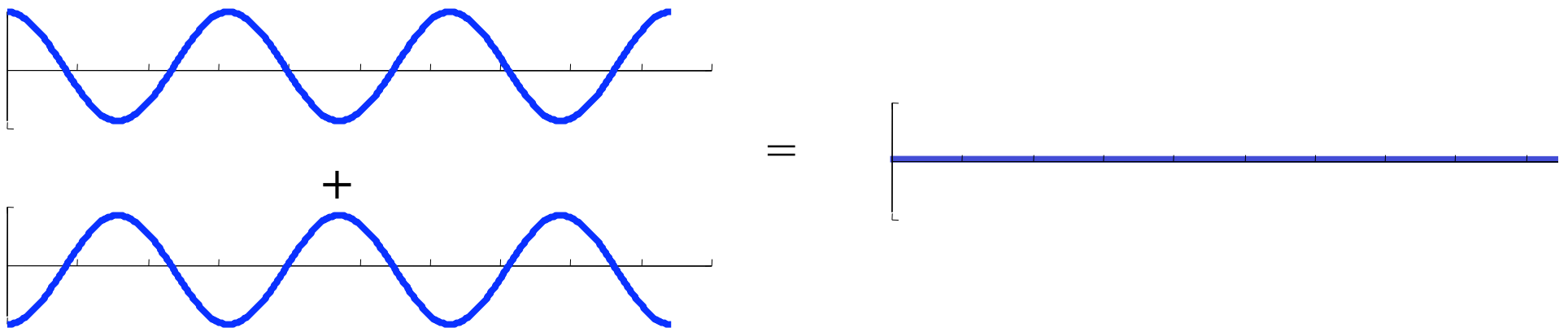
$$\Psi = \psi e^{i\phi}$$

↖
↖
amplitude *phase*

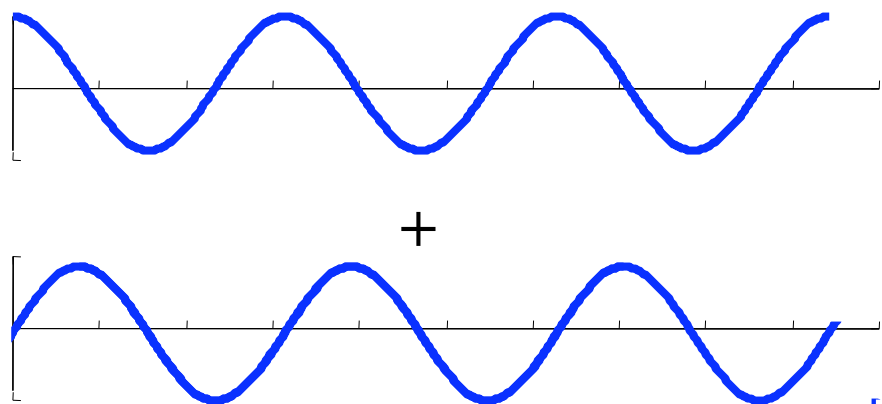
Waves: constructive interference



Waves: destructive interference



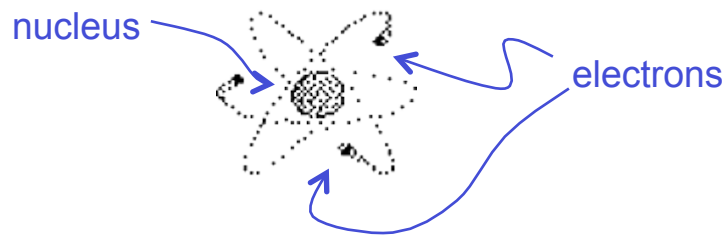
Test yourself!



?

- ①
- ②
- ③
- ④
- ⑤

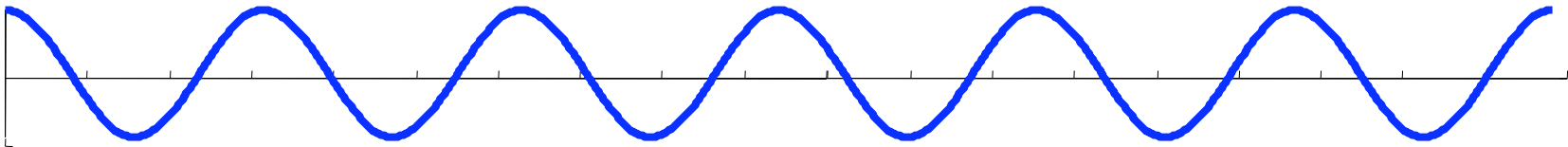
How does a wavelike electron lead to quantized orbitals in atoms?



The deBroglie hypothesis:

$$\lambda = \frac{h}{m_e v}$$

Imagine a wavelike electron wrapping many, many times around the nucleus:



The electron can only exist in orbits with no destructive interference!

Both the angular momentum and the energy are “quantized”

Quantum mechanics: wave-particle duality

Quantum mechanical view of the atom

Electrons are both waves and particles.

When detected, the electron appears to be a particle.

But the probability of finding the electron behaves like a wave.

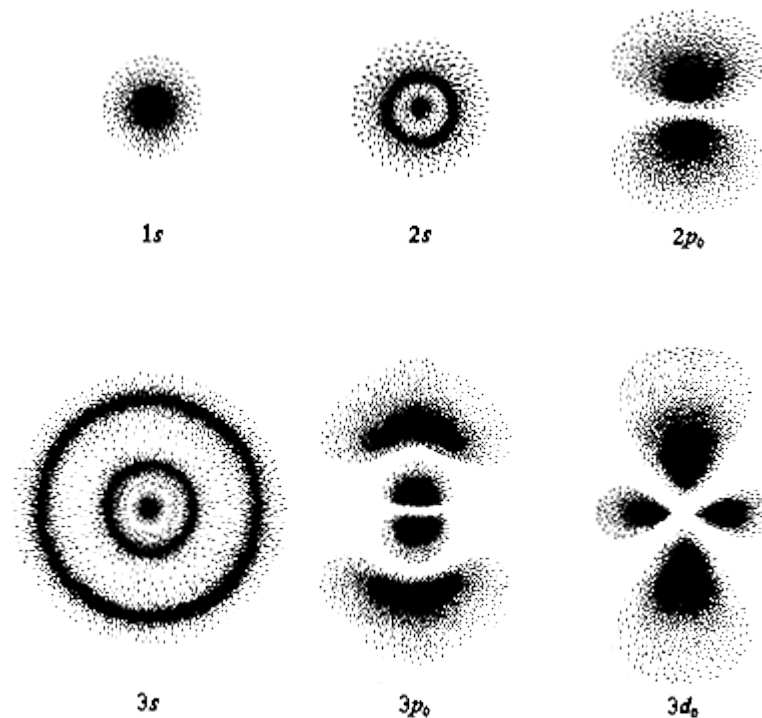


Figure 6-12. Probability density plots of some hydrogen atomic orbitals. The density of the dots represents the probability of finding the electron in that region.

© 1983 University Science Books; "Quantum Chemistry" by Donald A. McQuarrie



<http://www.scot-canoe.org/touring/graphics/photocomp/2002/ripples.JPG>

Young's double slit experiment: demonstration of wave-like nature of light

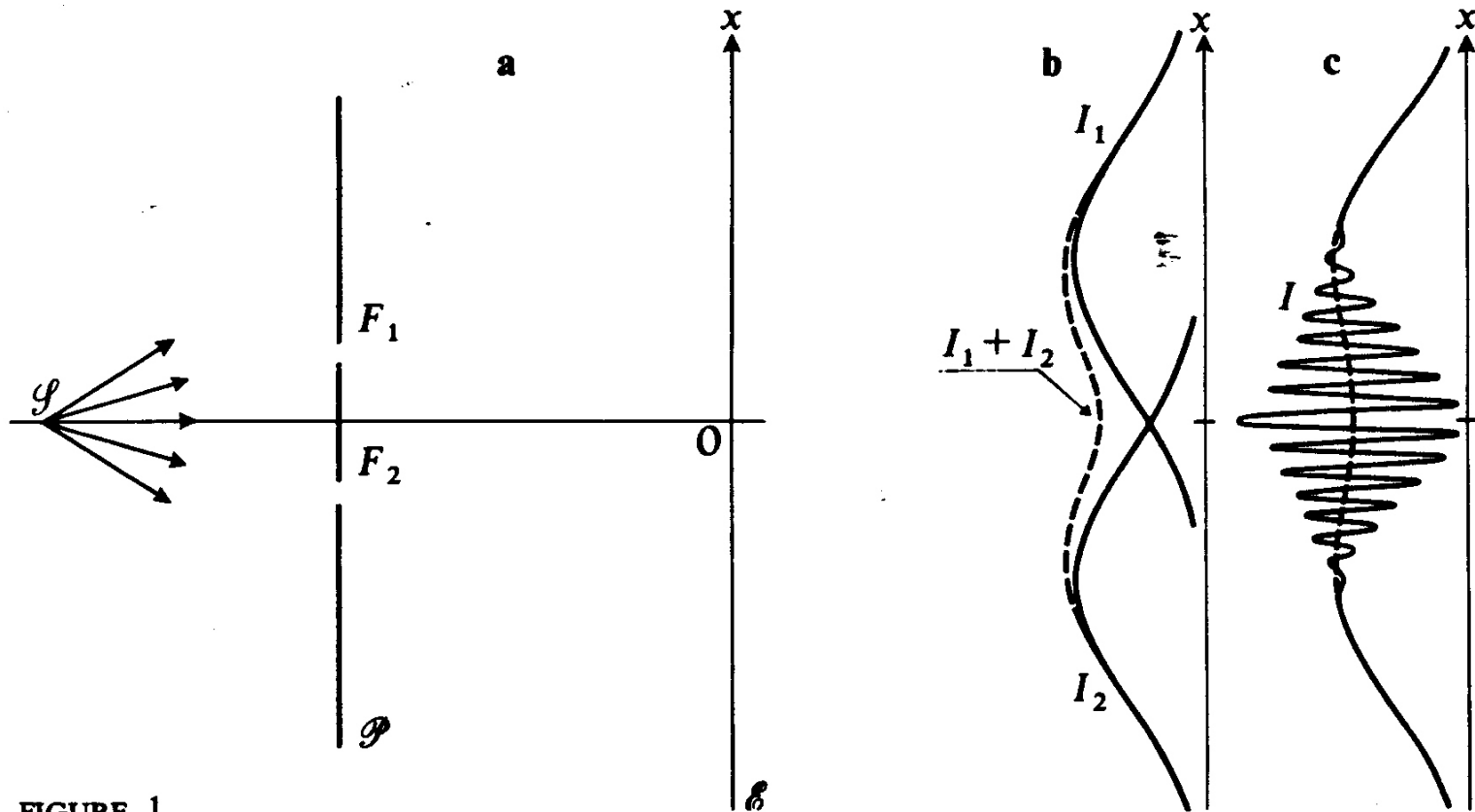


FIGURE 1

Diagram of Young's double-slit light interference experiment (fig. a). Each of the slits F_1 and F_2 produces a diffraction pattern on the screen \mathcal{P} . The corresponding intensities are $I_1(x)$ and $I_2(x)$ (solid lines in figure b). When the two slits F_1 and F_2 are open simultaneously, the intensity $I(x)$ observed on the screen is not the sum $I_1(x) + I_2(x)$ (dashed lines in figures b and c), but shows oscillations due to the interference between the electric fields radiated by F_1 and F_2 (solid line in figure c).

Double-slit experiment for electrons?

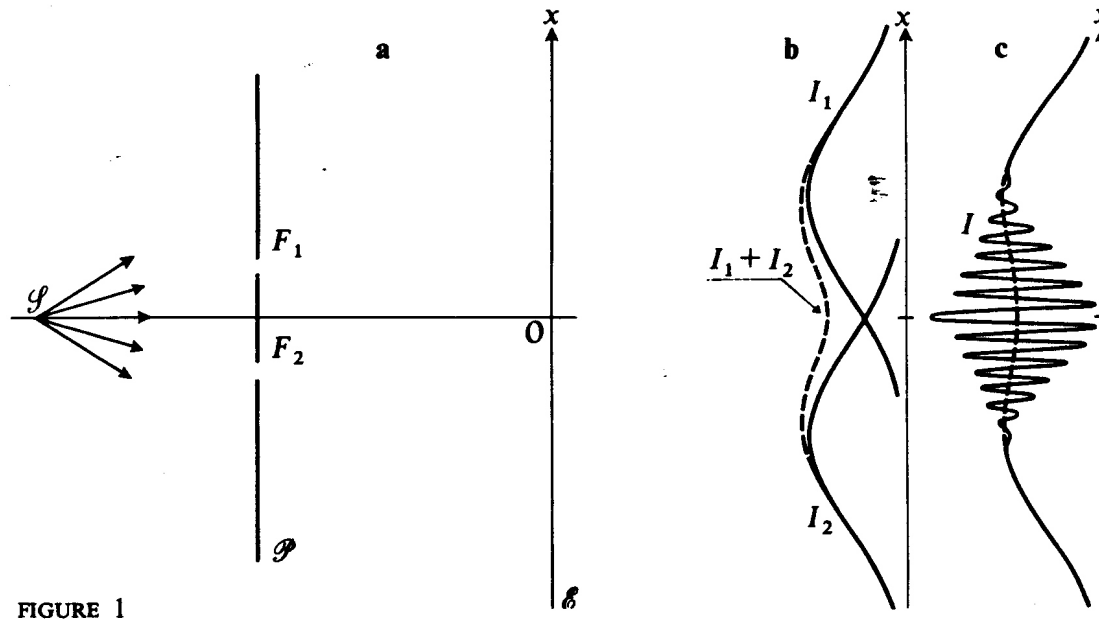
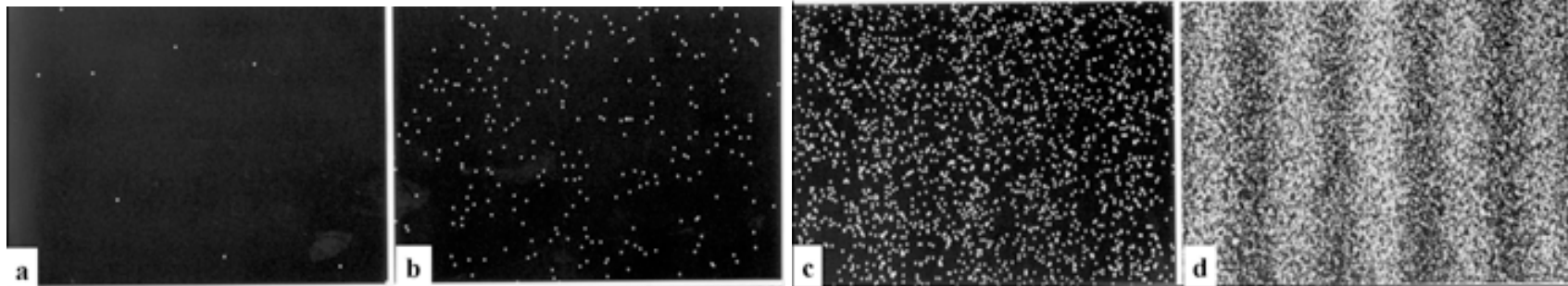
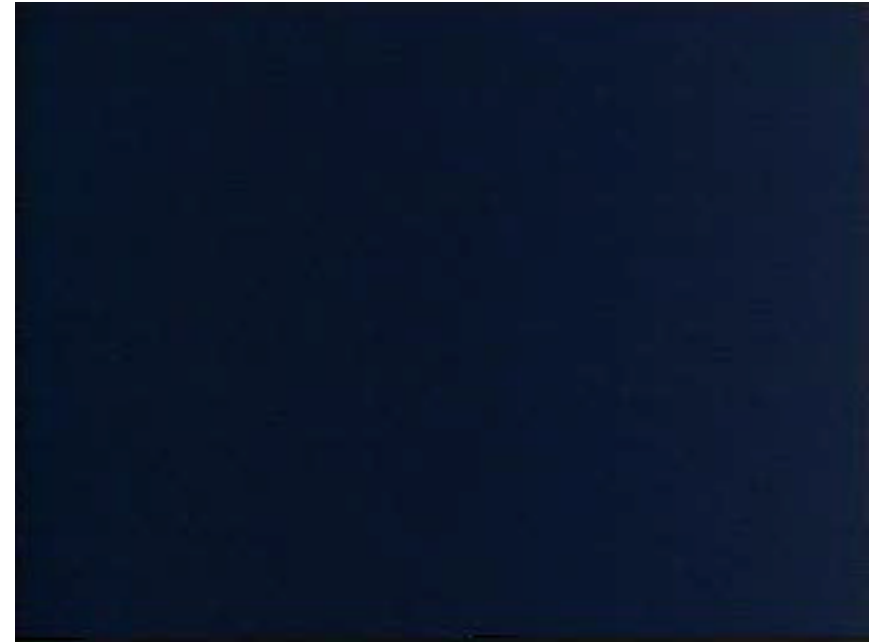
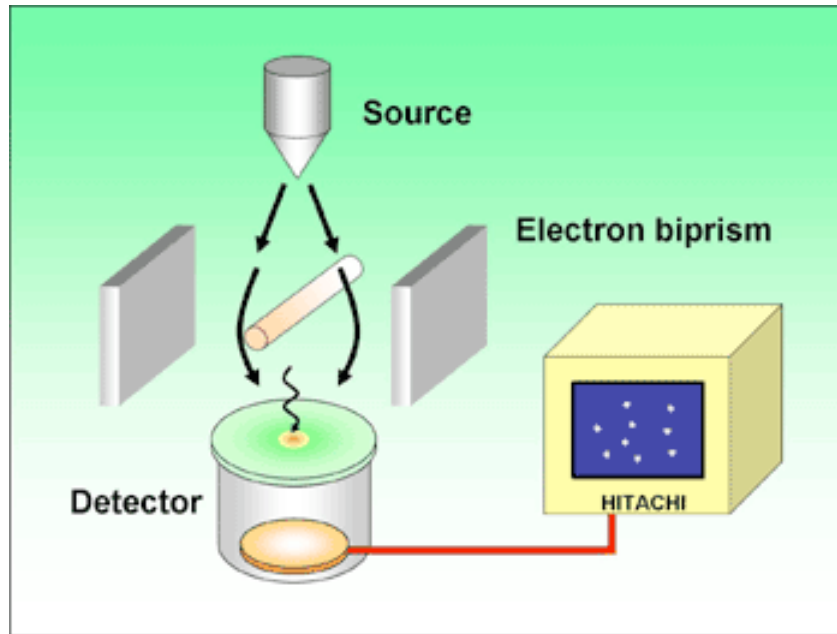


FIGURE 1

“We should say right away that you should not try to set up this experiment...The trouble is that the apparatus would have to be made on an impossibly small scale to show the effects we are interested in. We are doing a ‘thought experiment,’ which we have chosen because it is easy to think about.”

--Feynman Lectures on Physics, 1961-1963
(Volume III, Section 1-4)

Möllenstedt-Düker electron biprism experiment (Tonomura implementation)



Original experiments G Möllenstedt and H Düker 1955, 1956.

This data: <http://www.hqrd.hitachi.co.jp/em/doubleslit.cfm>, A. Tonomura, J. Endo, T. Matsuda, T. Kawasaki and H. Ezawa, "Demonstration of Single-Electron Buildup of an Interference Pattern" Amer. J. Phys. 57 (1989) 117-120.

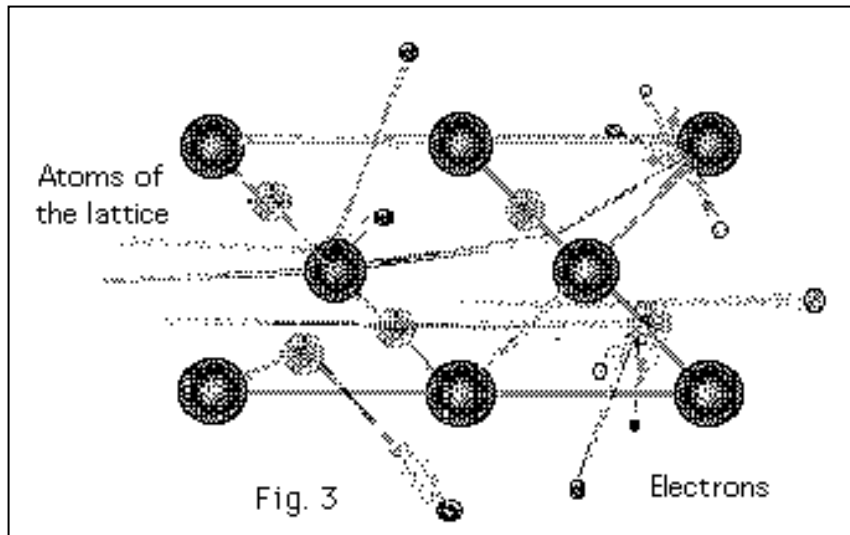
Why is wavelike electron behavior not obvious in every day metals and semiconductors?

Electrons typically have small wavelengths.

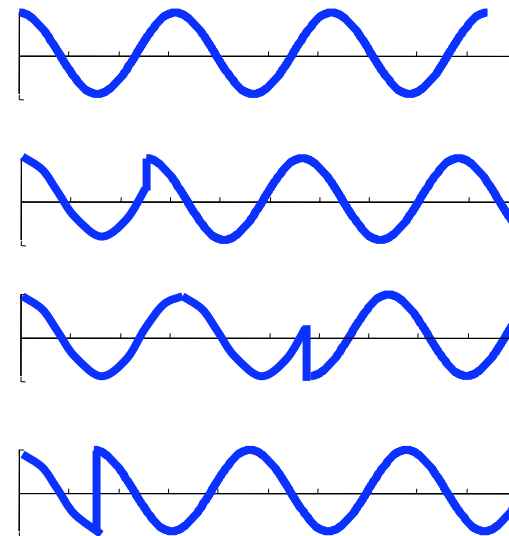
Electrons with different energies have different wavelengths.

Electrons lose phase coherence in various ways.

Artist's rendition of electrons colliding with lattice from Oak Ridge National Lab

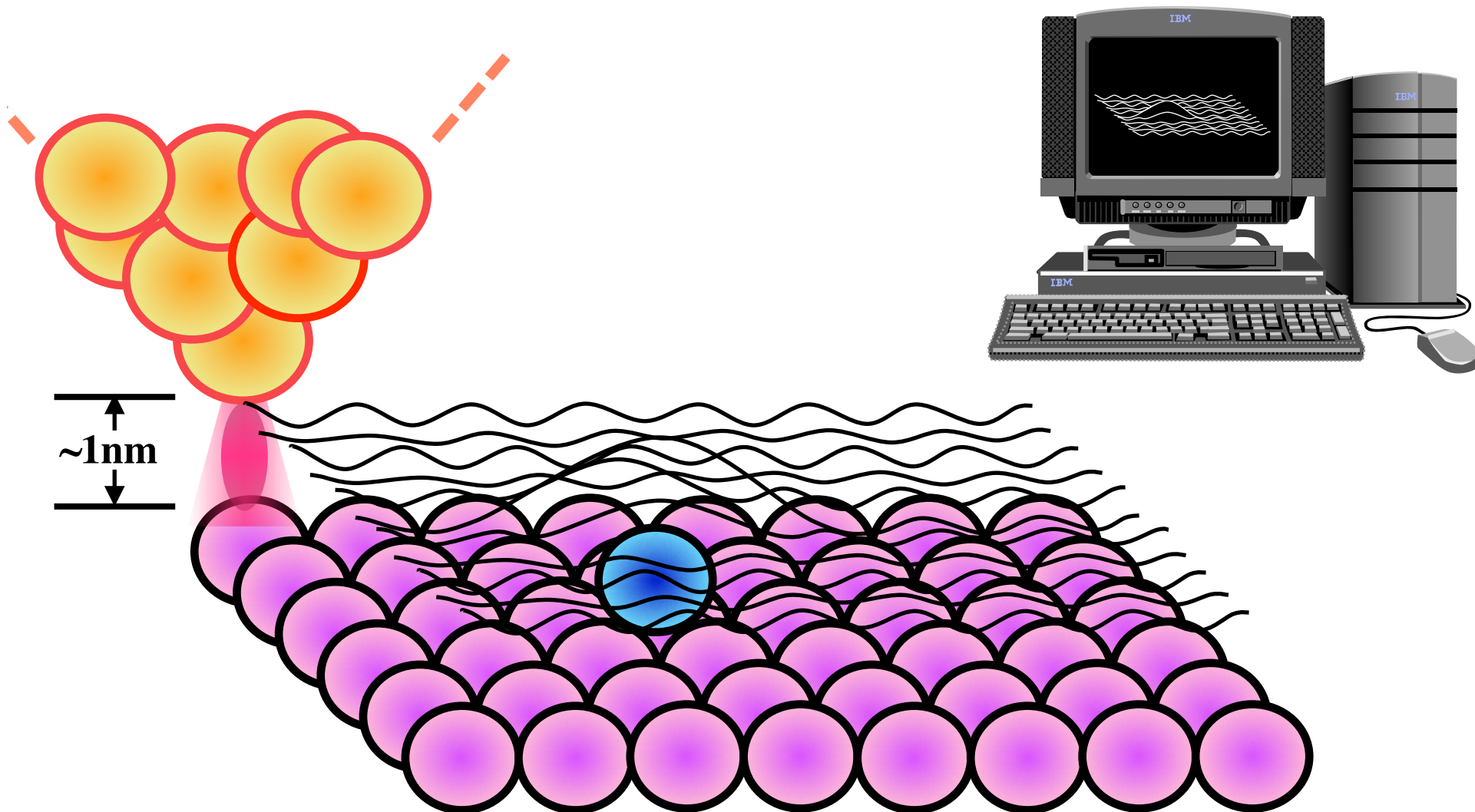


decoherence/dephasing

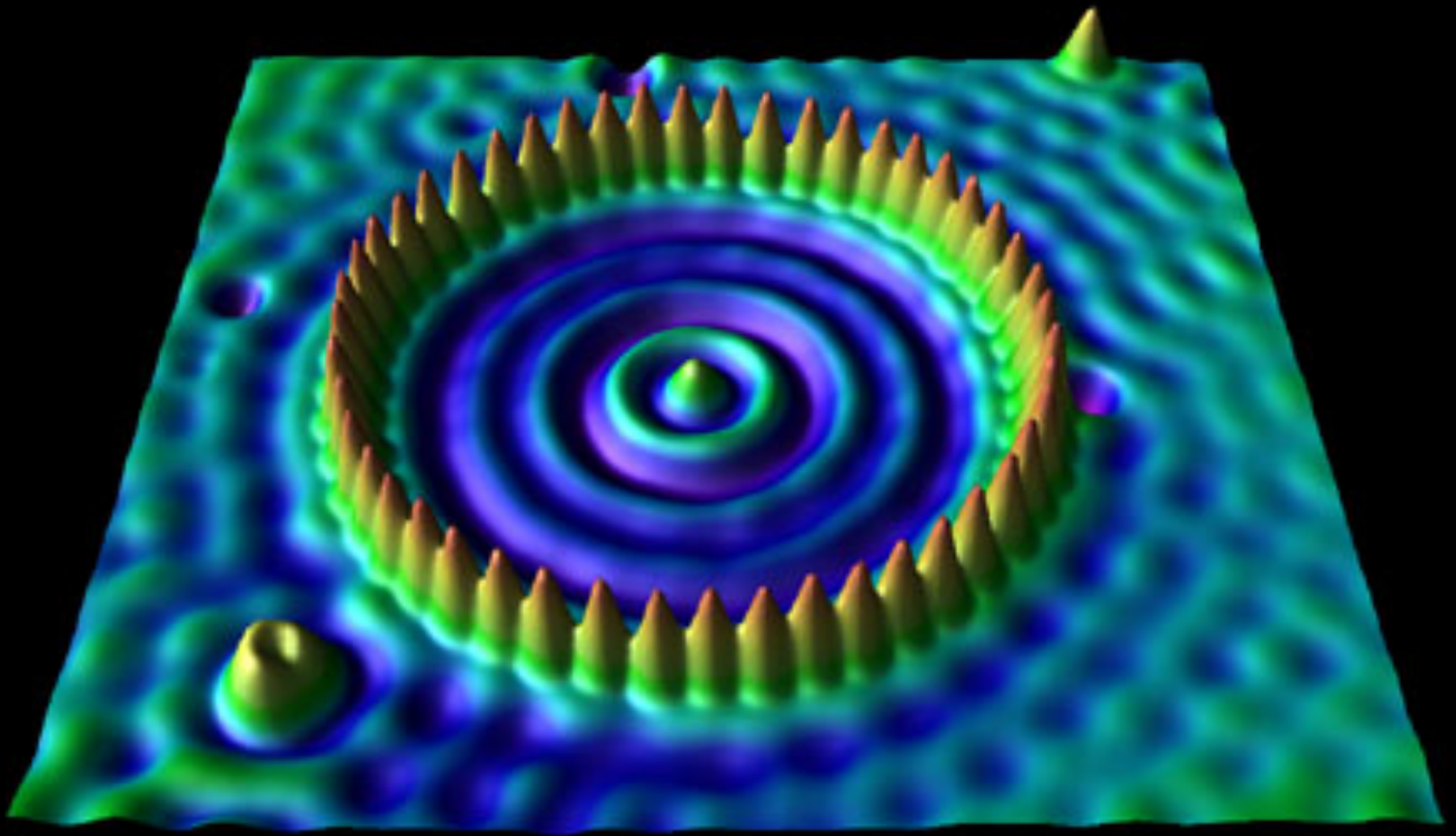


How can we see electron waves in real materials?

Scanning Probe Microscopy



Source: Don Eigler, Barbara Jones



“Fe on Cu(111)”
Crommie, Lutz, Eigler (1993)

Outline

Introduction to Quantum Mechanics

Wavelike electrons save the atom

Show me a real wave!

Why we don't usually experience wavelike/quantized behavior: Decoherence

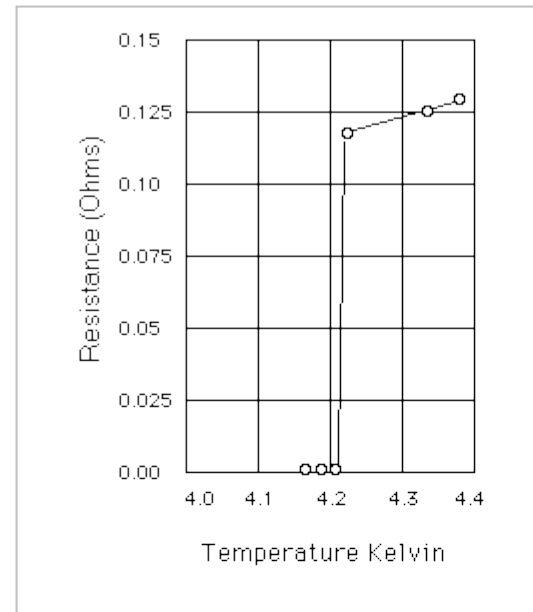
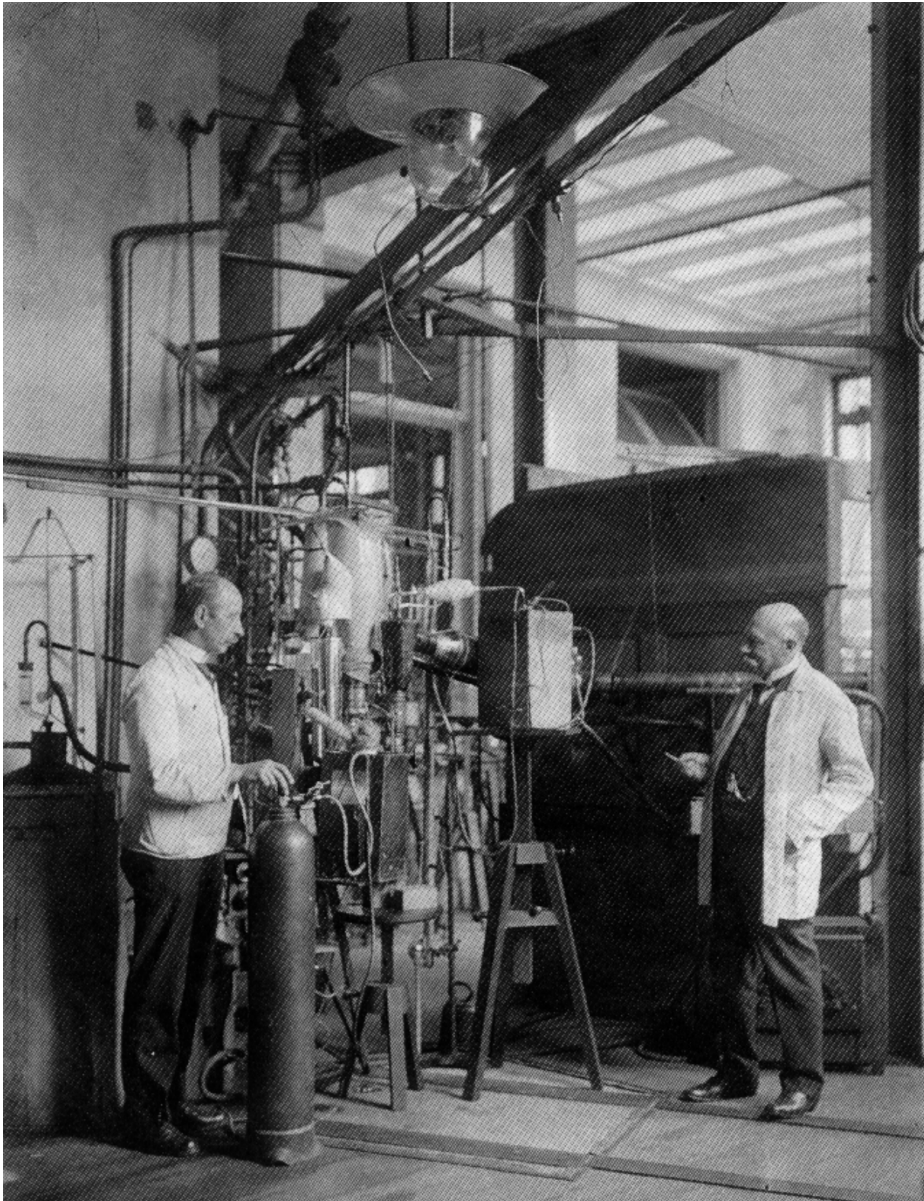
To see wavelike behavior in normal metals and semiconductors, we usually need to look at very small length scales (the nanoscale)

Quantum Whirlpools

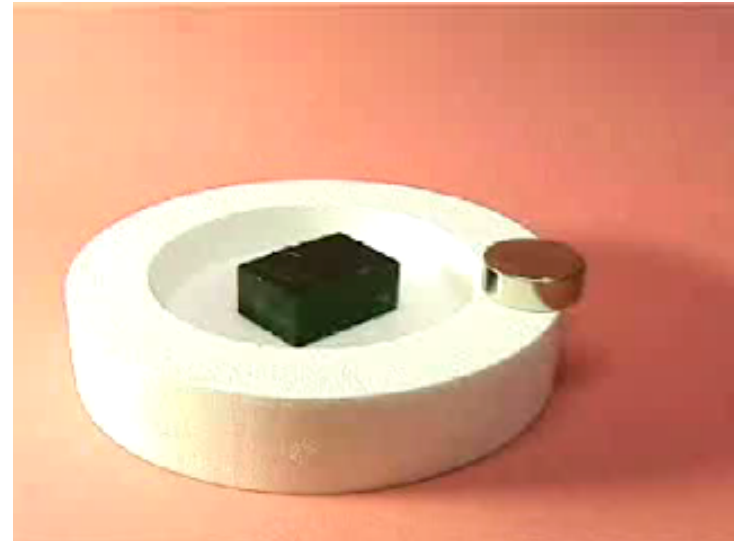
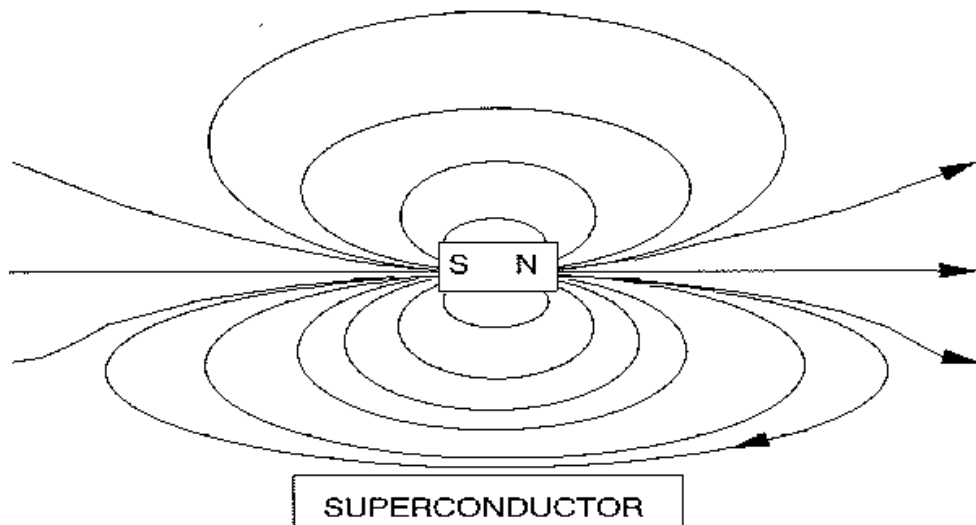
Superconductivity: A charged macroscopic quantum fluid

Vortices in superconductors

Discovery of Superconductivity 1911



Meissner Effect (1933) and Levitation



Note: complete analysis of levitation effect requires not only gravity and the Meissner effect, but also vortex pinning

Image at left:

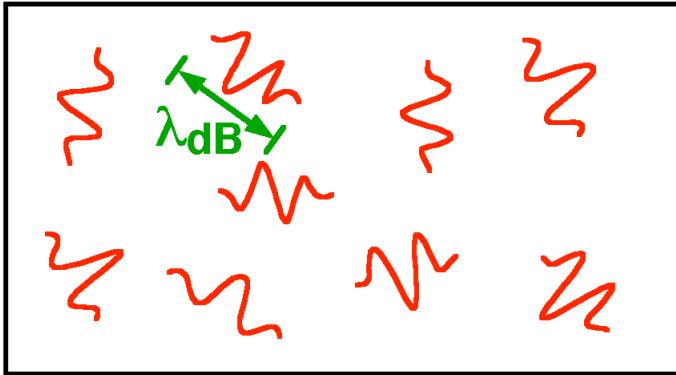
<http://www.ornl.gov/info/reports/m/ornlm3063r1/pt2.html>

Movie at right:

Original source unknown to me, widely available on web

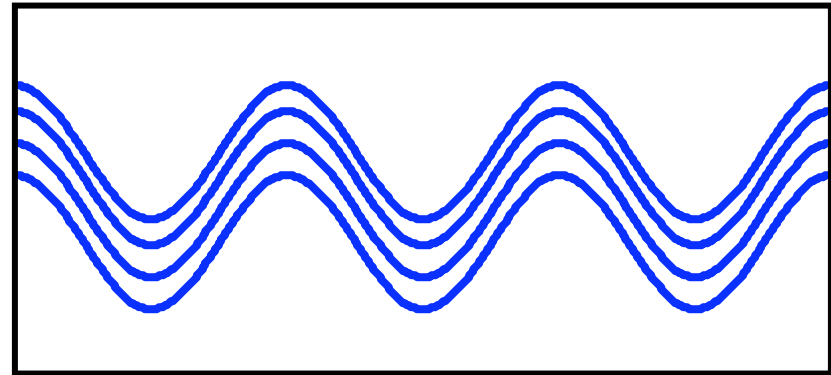
Superconductor = Macroscopic Quantum Superfluid that carries charge

Ordinary Gas or Fluid



particles move somewhat independently
many small waves
doesn't look wavelike over large distances

Quantum "Superfluid"



particles "march in lockstep"
one big wave made of many particles
coherent, wavelike behavior over large distances

Superconductors are superfluids made of *charged* particles.

Aside: the phrase "macroscopic quantum" has multiple uses.
A "macroscopic quantum state" has long-range (macroscopic) coherence.
However, the phrase "macroscopic quantum coherence" is usually reserved for Schrödinger cat states.

Vortices

Vortices

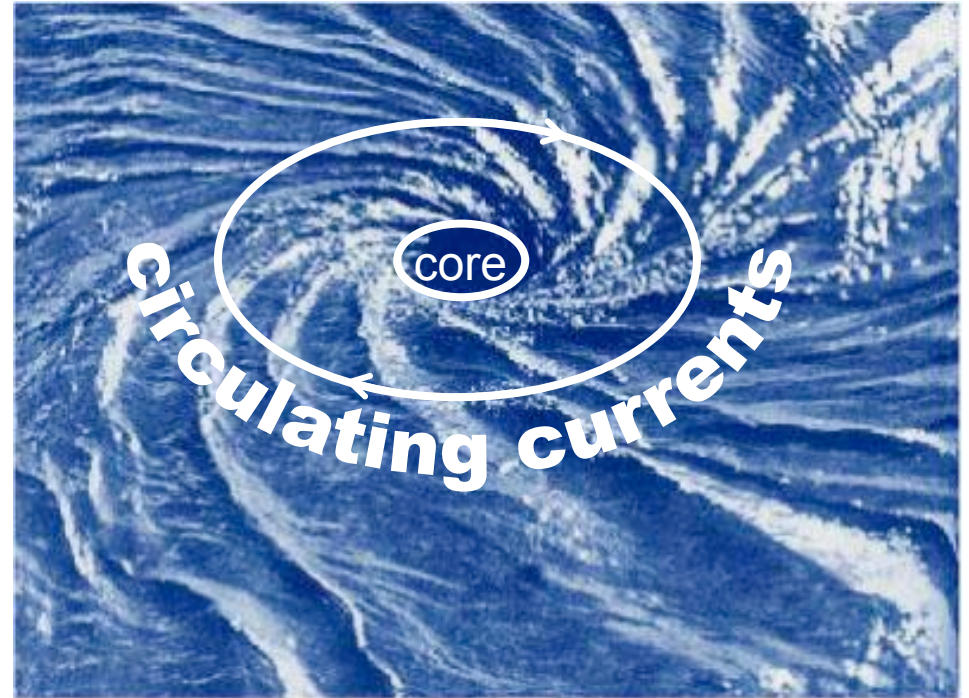
- cylindrical symmetry
- fluid swirling around a core

Vortices in Quantum Fluids

- persistent
- quantized

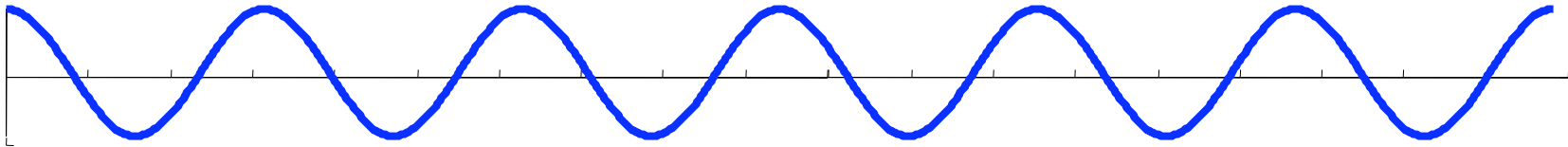
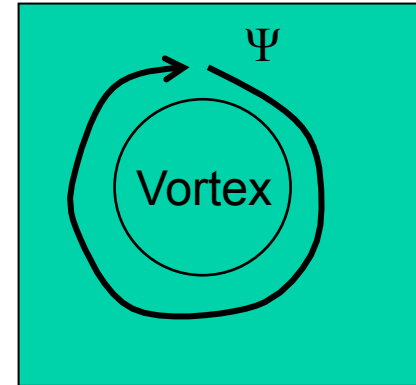
Vortices in Superconductors

- generate a local magnetic field



A Quantum Vortex Is *Quantized*

The “pseudowavefunction” is a wave. If you imagine wrapping it around the vortex core, the phase can only change by multiples of 2π



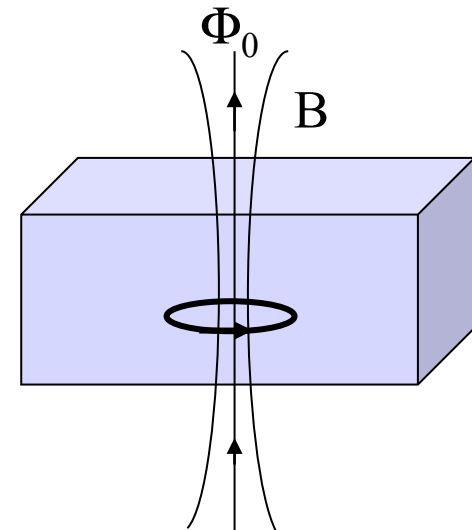
In the lowest energy configuration, it changes by exactly 2π

This leads to the quantization of magnetic flux:

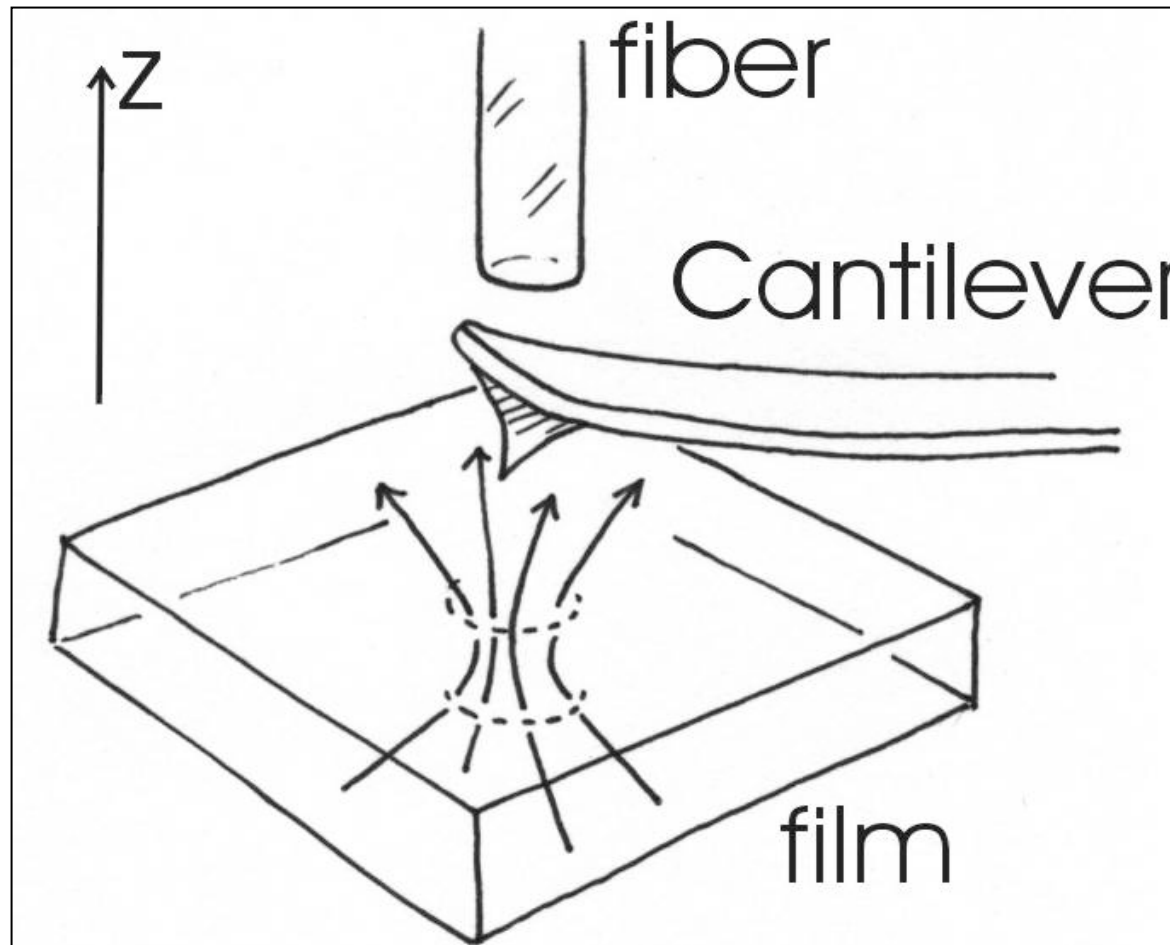
$$\Phi = \int \vec{B} \cdot d\vec{a} = n\Phi_0$$

Where n is an integer
and the superconducting flux quantum

$$\Phi_0 = hc/2e = 20.7 \text{ Gauss } \mu\text{m}^2$$

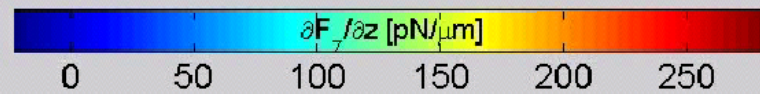
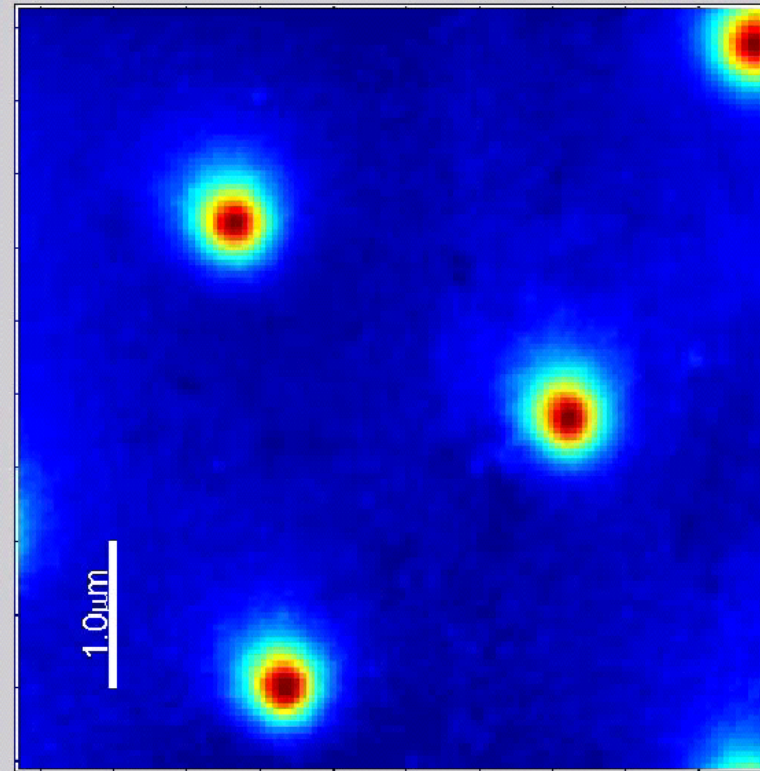
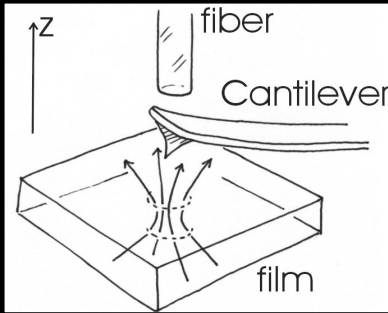


“Seeing” vortices with Magnetic Force Microscopy



Other ways to see vortices: Bitter decoration, scanning tunneling microscopy, magneto-optics, Hall probe microscopy, SQUID microscopy.

Experimental Data: Images of Vortices in Superconducting $\text{YBa}_2\text{Cu}_3\text{O}_{6.99}$



Why do we care about vortices in superconductors?

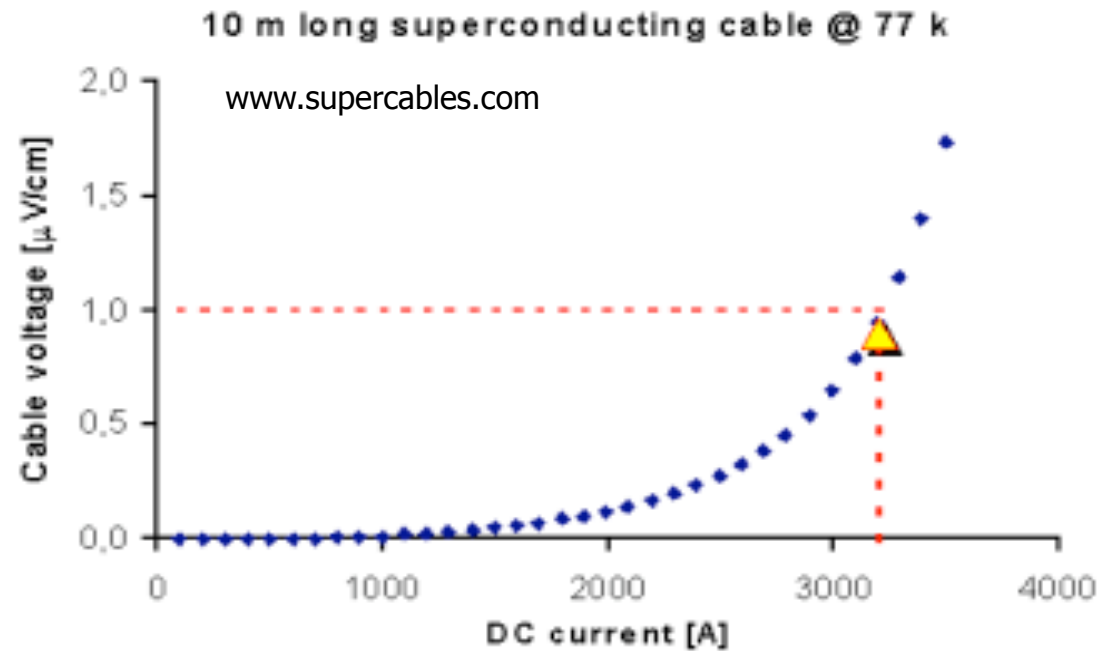
- 1) Fundamental interest
- 2) Superconducting technology

Large classical vortices in air cause undesirable dissipation.



Nanoscale quantum vortices in superconductors cause undesirable dissipation too.

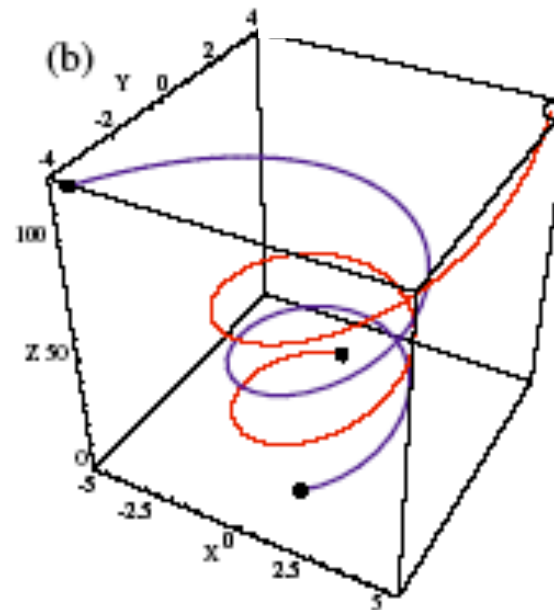
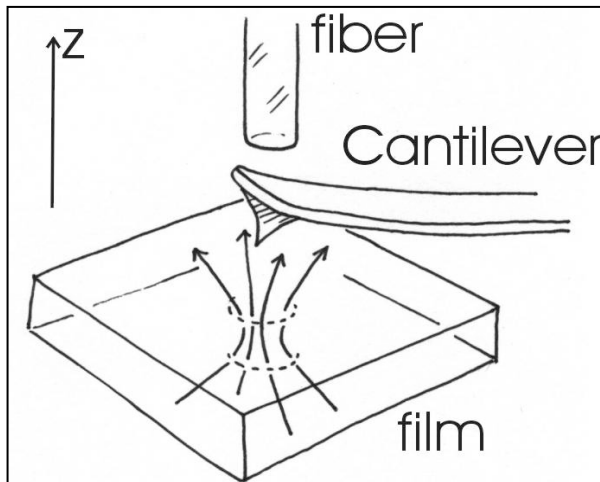
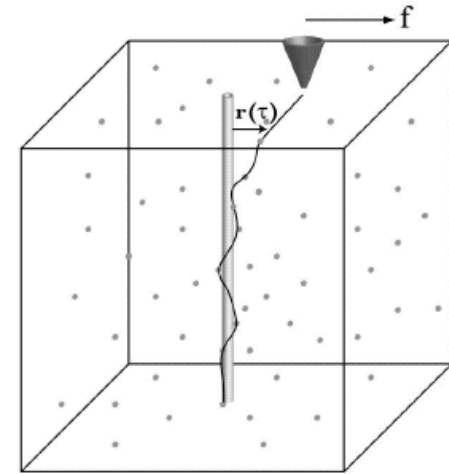
Currents make the vortices flow and destroy the perfect conductivity that makes superconductors so useful.

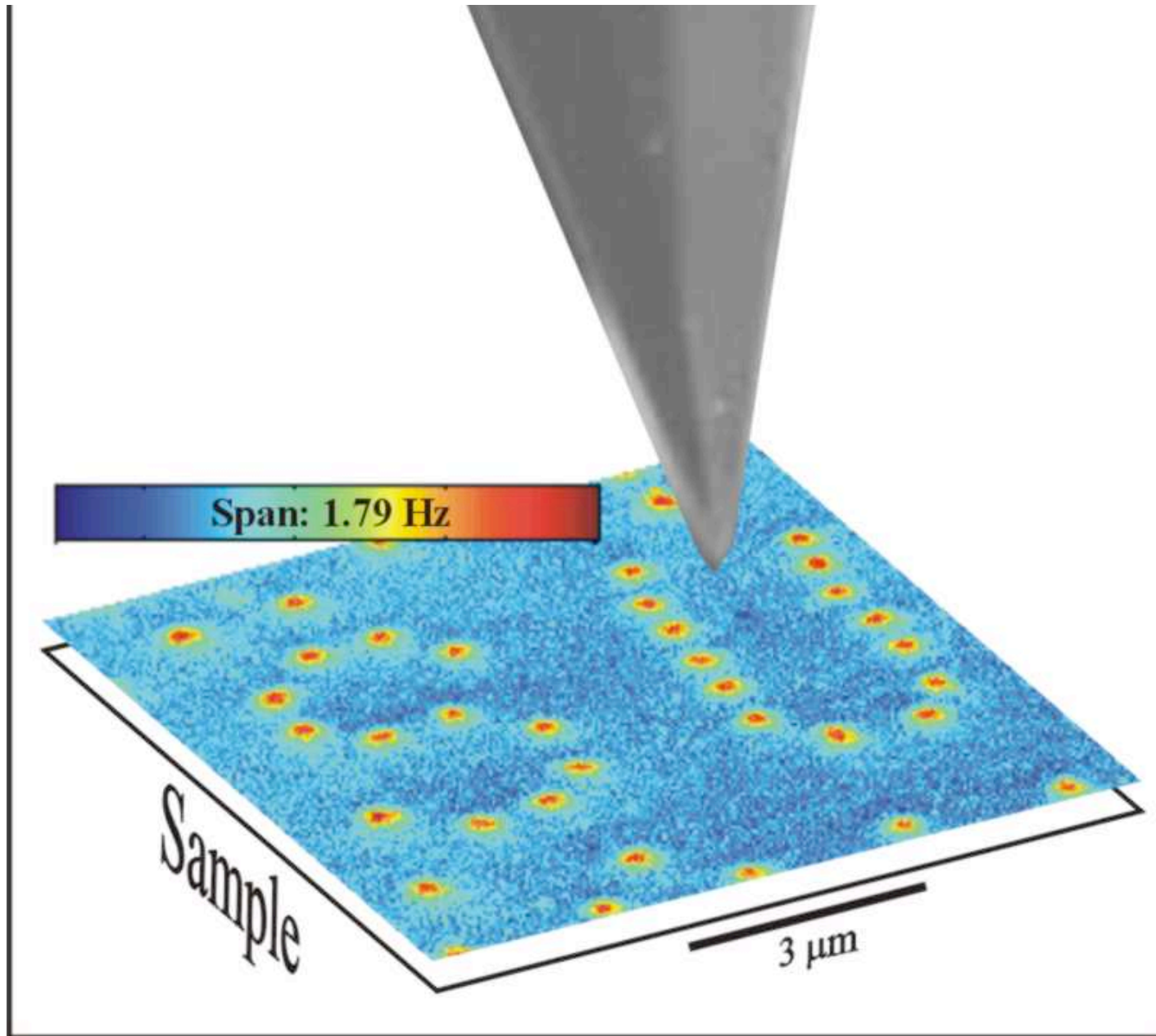


Single-vortex manipulation

To address questions such as:

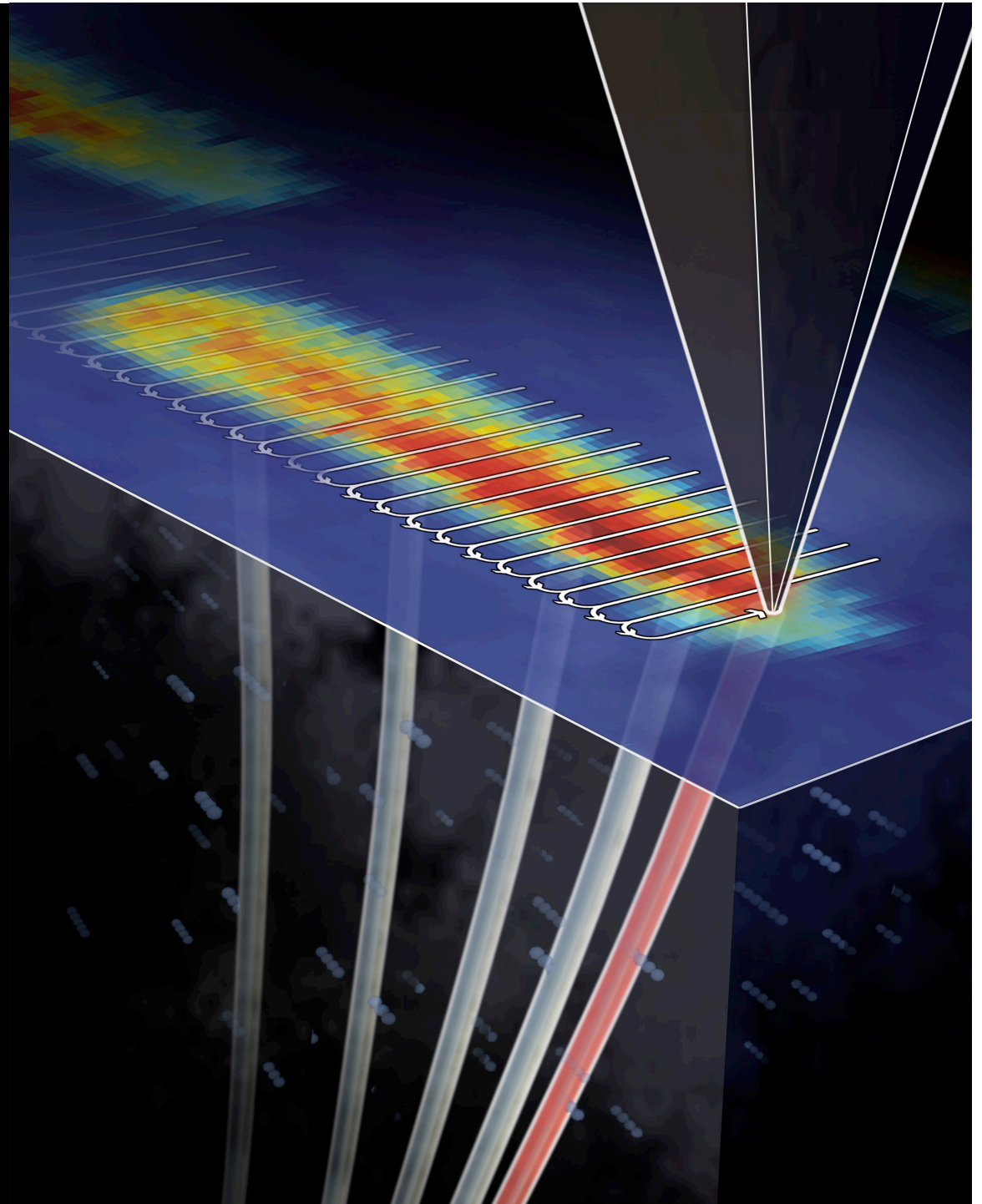
- How do vortices interact with the crystal lattice and its defects?
- How do vortices tilt?
- Do vortices entangle?





Dragging quantum vortices in niobium thin films:
Motion detected/controlled to better than 10 nm

Dragging quantum vortices in superconducting YBCO:
The vortex acts like a string moving through a landscape of defects that pin it.



Credits

Images

Todd's Quantum Intro Web Page
IBM (Dr. Don Eigler)
Scottish Canoe Association
Oak Ridge National Laboratory
Edubuzz.org
Clifford Hicks
David Pritchard
Zlatko Tesanovic
Superconductor Lab Oslo

Vortex Coworkers

Ophir Auslaener
Lan Luan
Jenny Hoffman
Eric Straver
Ruixiang Liang
Walter Hardy
Doug Bonn
John Kirtley
Martin Huber

Funding

DOE BES
AFOSR
NSF
IBSF



Download Hands-on Nano Activities from the Stanford Center for Probing the Nanoscale:

- <http://teachers.stanford.edu/activities/>

Summer Institute for Middle School Teachers

at the Stanford CPN:

- <http://www.stanford.edu/group/cpn/education/>



Explore Nano further at <http://teachers.stanford.edu/links/>

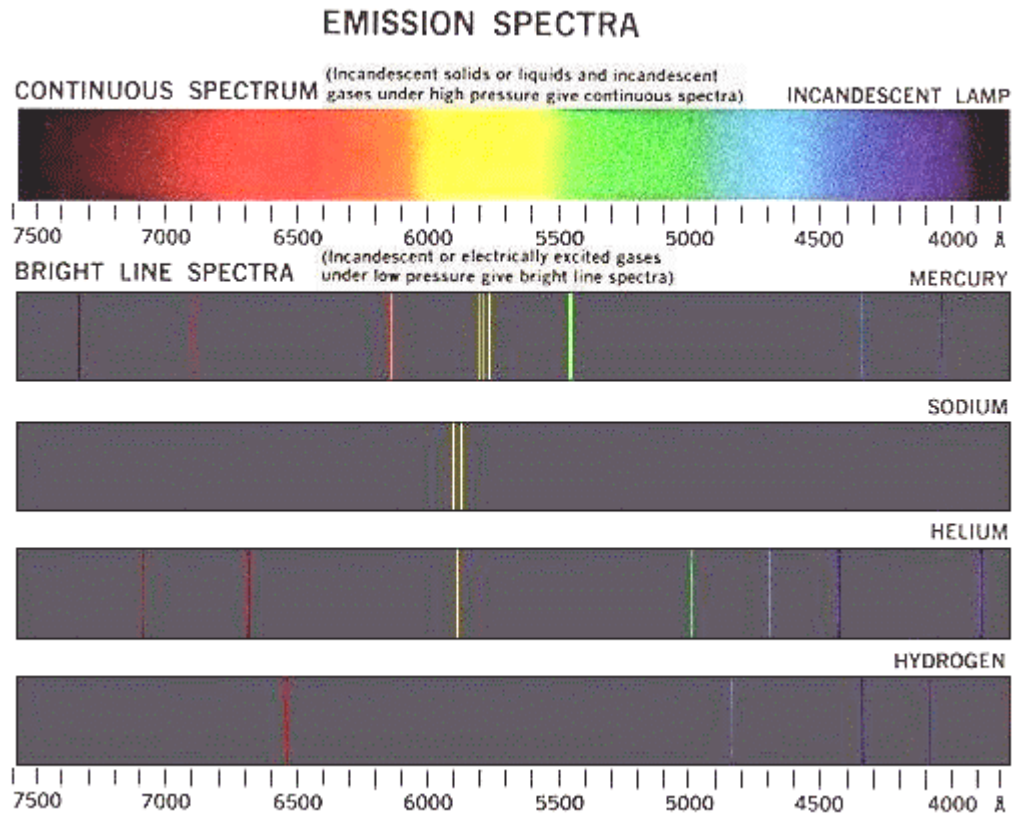
- Nano Games (Activities from University at Albany)
- NanoSense (Activities)
- NISEnetwork (Activities, media)
- nanoHUB (Presentations, media)
- NanoKids (Activities, presentations from RICE University)
- Resource Area For Teachers (hands-on activities and supplies)

for more info, contact

kmoler@stanford.edu or Tobi Beetz, tobi@stanford.edu

Extra Slides

Quantized atomic orbitals



continuous spectrum
of visible light

mercury (quantized)

sodium (quantized)

helium (quantized)

hydrogen (quantized)

Quantum mechanics: wave-particle duality

Quantum mechanical view of the atom

Classical view of the atom

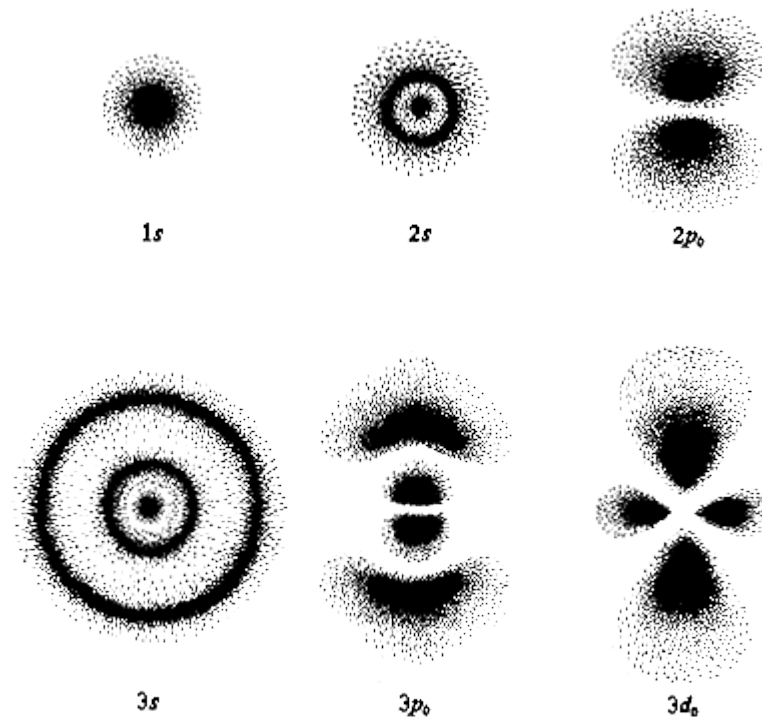


Figure 6-12. Probability density plots of some hydrogen atomic orbitals. The density of the dots represents the probability of finding the electron in that region.

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Quantum mechanics: wave-particle duality

Quantum mechanical view of the atom

Schrödinger's Equation

$$i\hbar \frac{\partial}{\partial t} \psi(\vec{r}, t) = \frac{\hbar^2}{2m} \nabla^2 \psi(\vec{r}, t) + V(\vec{r}, t) \psi(\vec{r}, t)$$

i is the imaginary number, $\sqrt{-1}$

\hbar is Planck's constant (divided by 2π)

m is the mass of the particle

∇^2 is the Laplacian operator,

$$\nabla^2 = \frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}$$

$V(\vec{r}, t)$ is the external potential

$\psi(\vec{r}, t)$ is the wave function

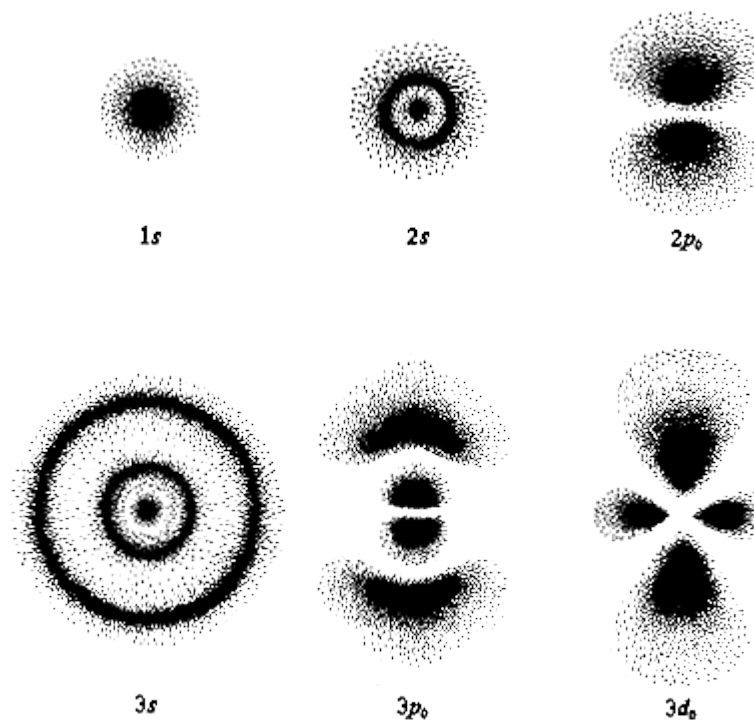


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Quantum mechanics: wave-particle duality

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The wave function

$$\psi(r, t)$$

$$|\psi|^2 \sim \textit{probability}$$

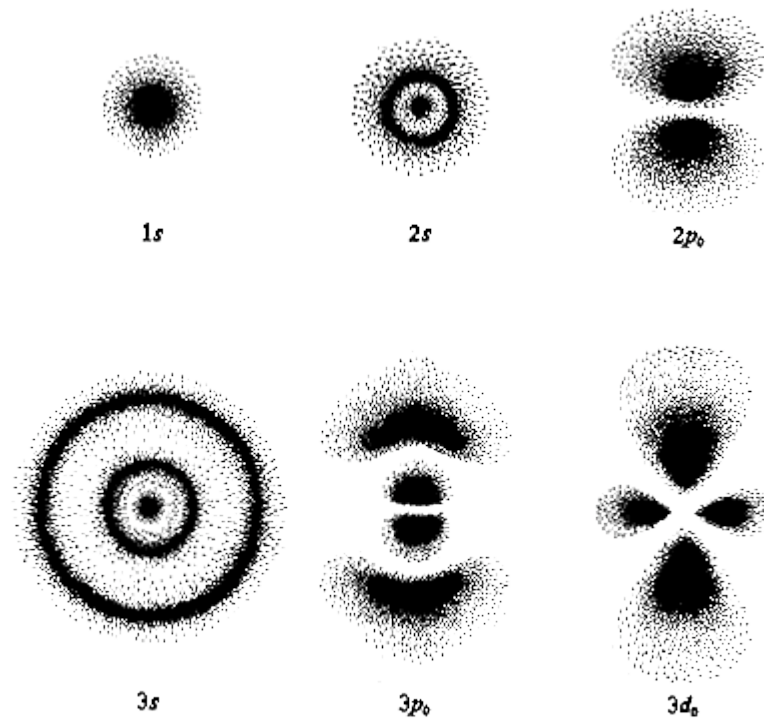


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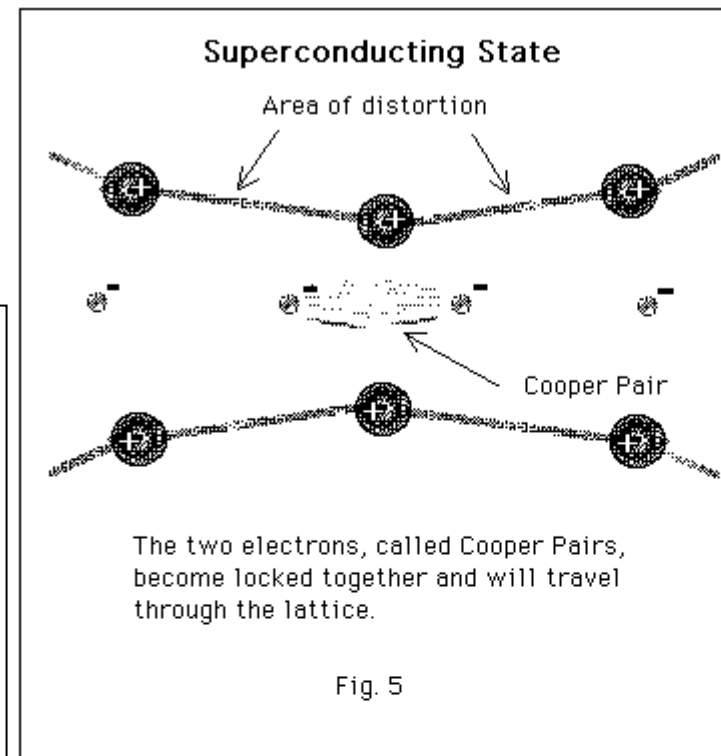
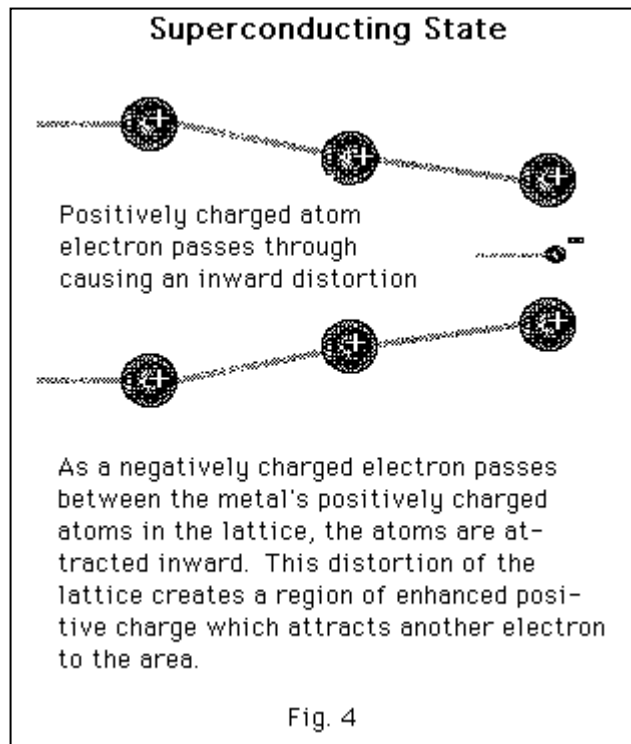
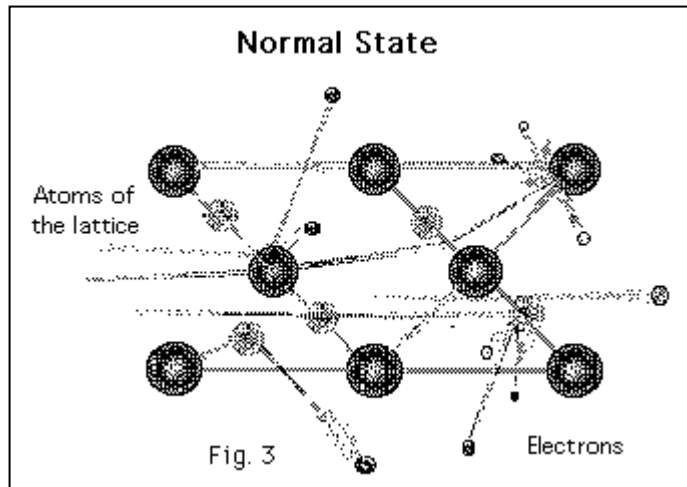
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Macroscopic Metaphor: “Drafting”



BCS Theory of Superconductivity 1957

phonon-mediated electron-electron interaction

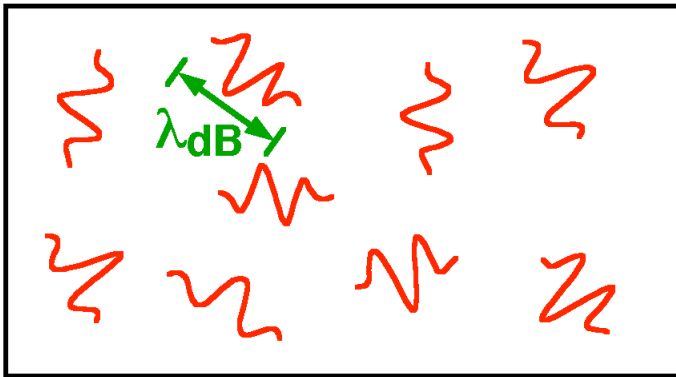


Like toddlers pair on the Galilee,
Electrons pair on a Fermi sea,
Superconductivity.



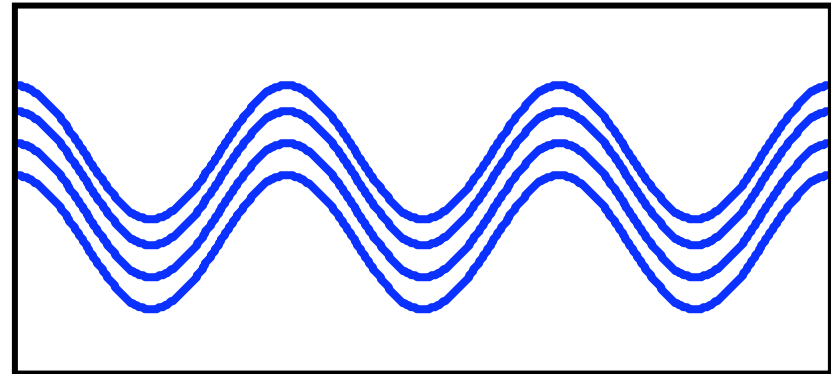
Macroscopic Quantum Fluids

Ordinary Gas or Fluid



particles move somewhat independently
many small waves
doesn't look wavelike over large distances

Quantum "Superfluid"



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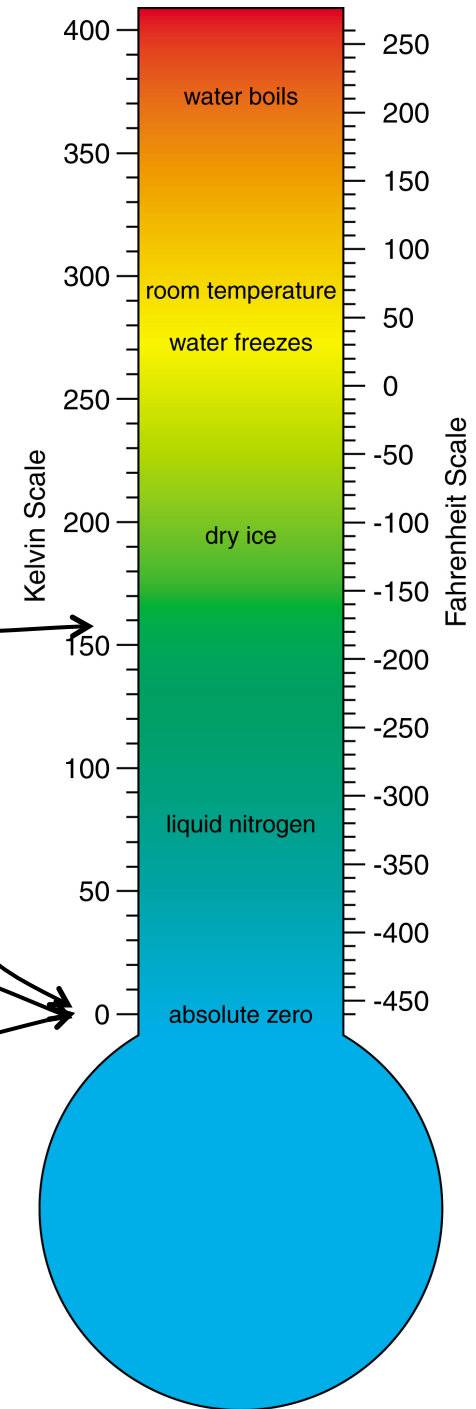
Temperature

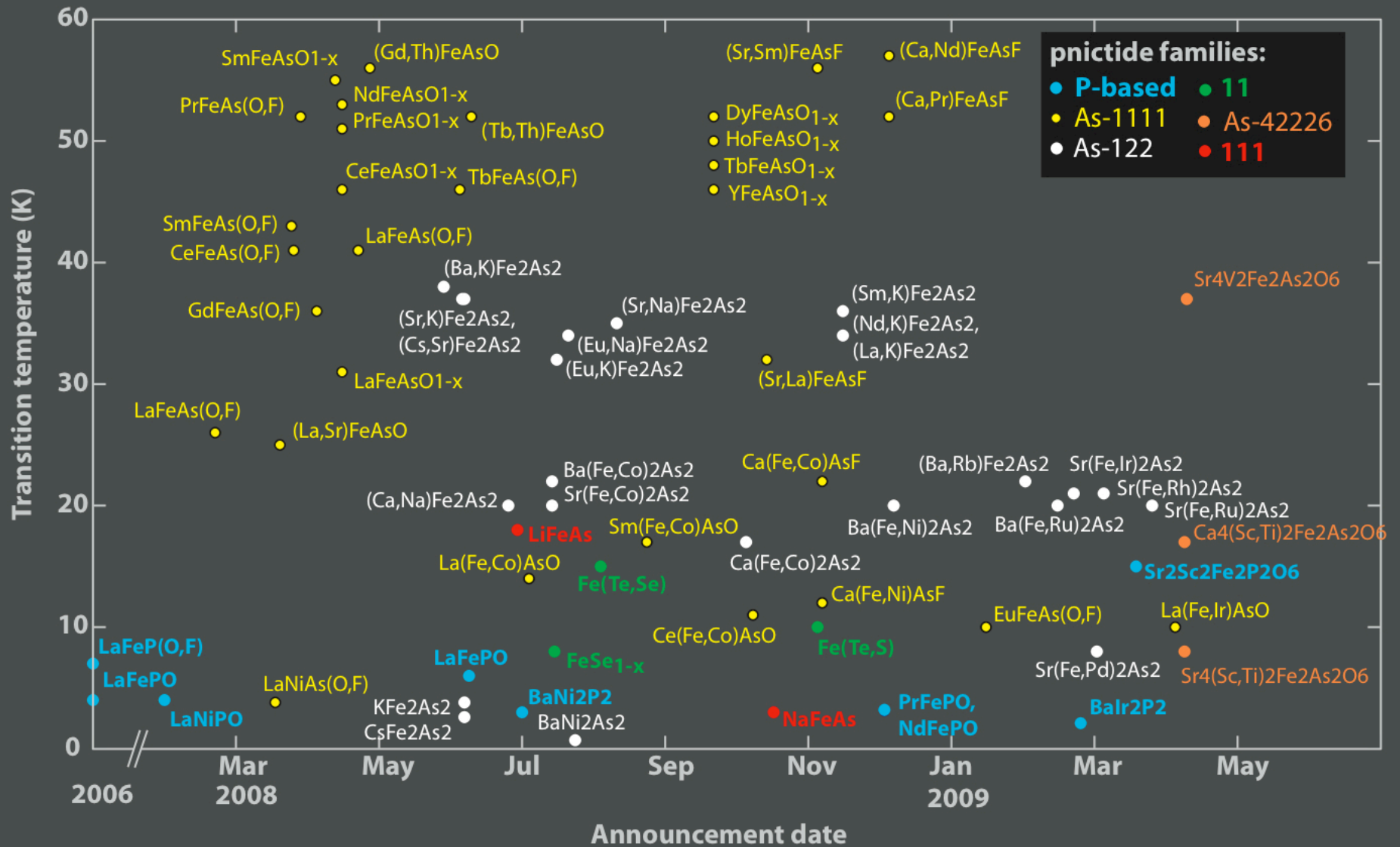
160 Kelvin
highest (known) temperature
superconductors
(excluding USO's)

2.2 Kelvin
superfluid He-4

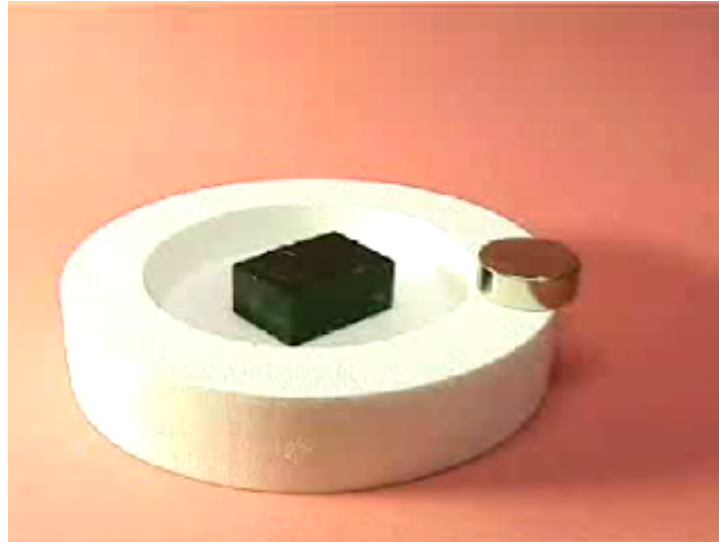
2 milliKelvin
superfluid He-3

sub-microKelvin
atomic Bose-Einstein
condensates



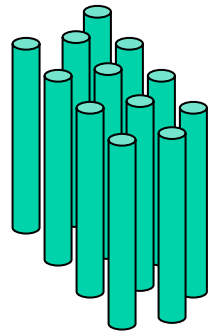


Levitation

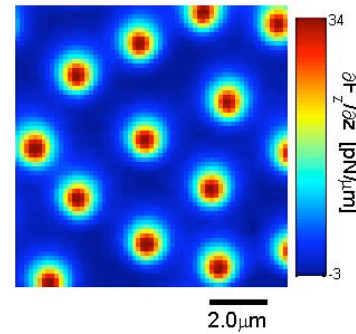


Visualization

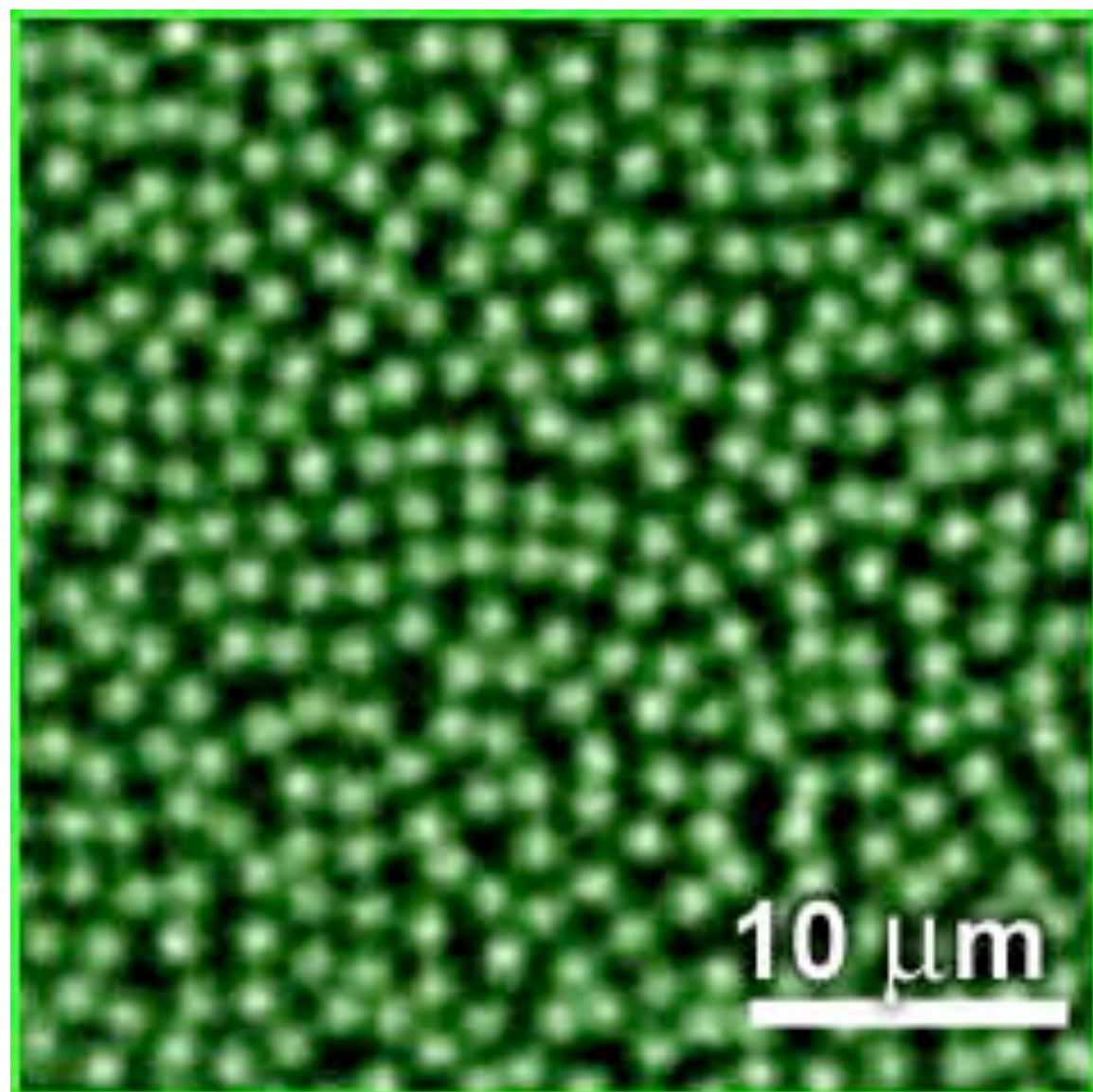
Vortices, each carrying 1 magnetic flux quantum, penetrate the superconductor.

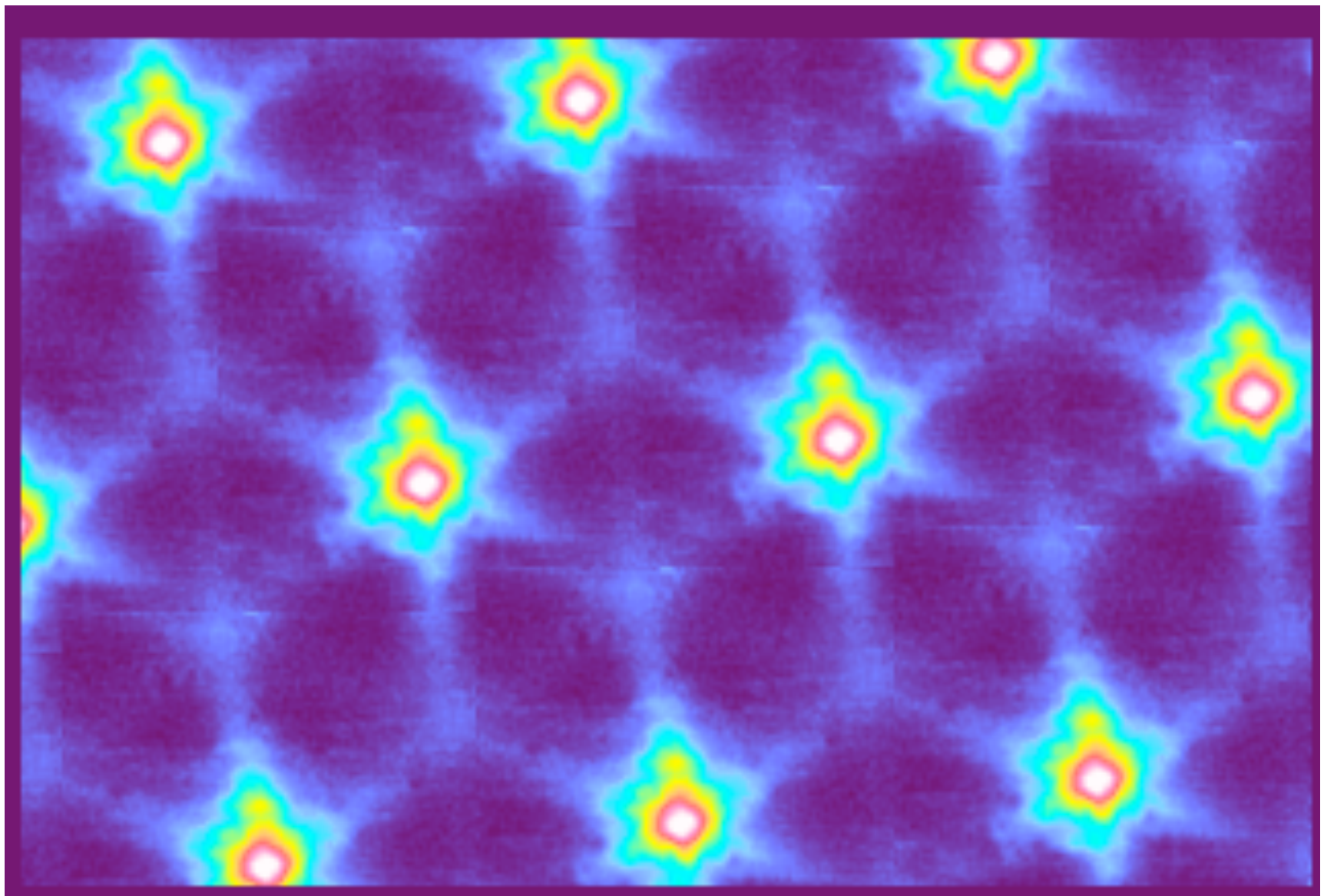


Sketch of vortices



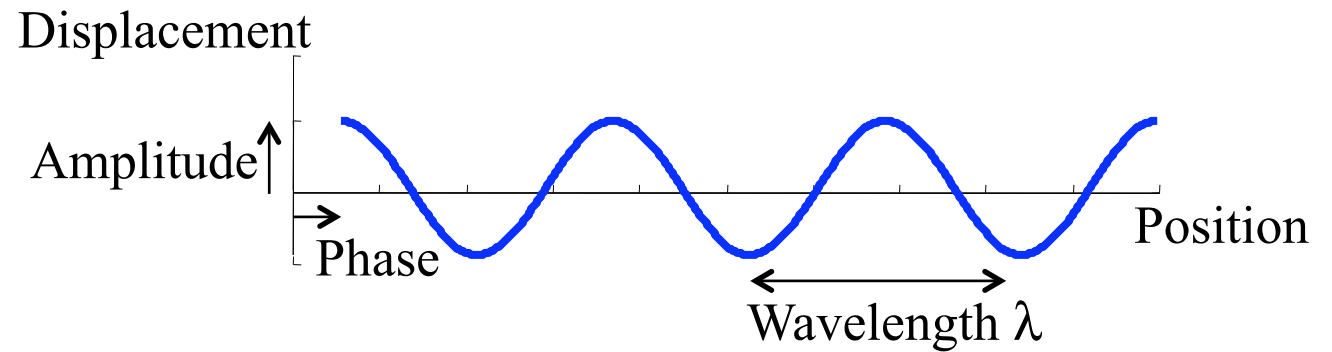
Actual magnetic image of YBCO surface



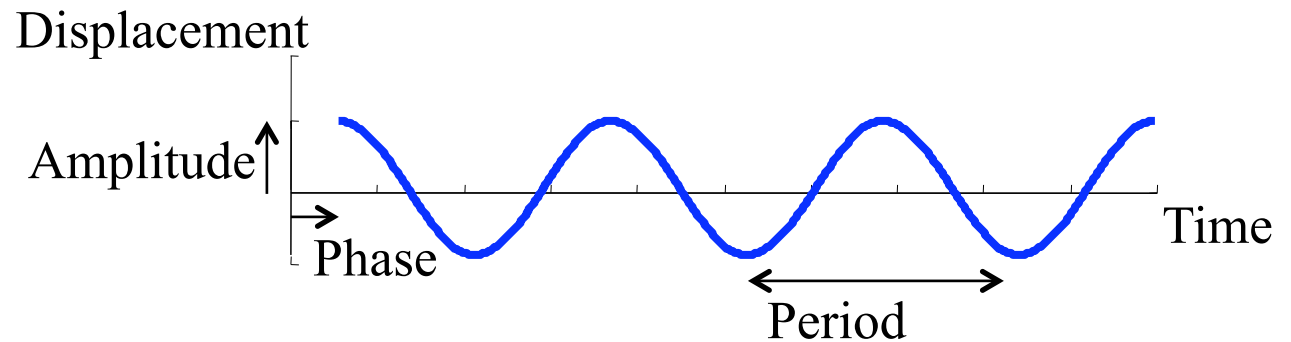


Mathematical example of a wave

At a fixed
time:



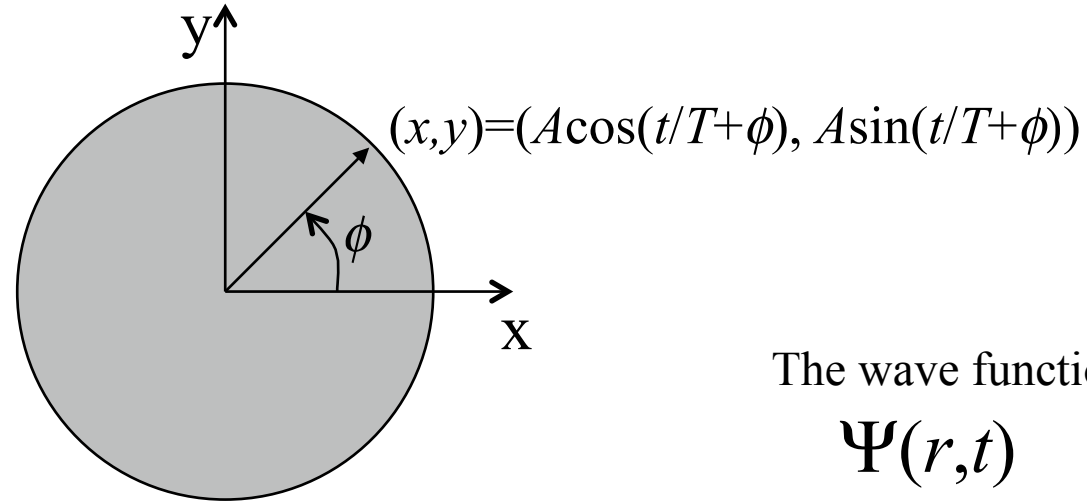
At a fixed
position:



$$\text{wavelength} = \text{velocity} \times \text{period}$$

Aside: *internal phase*

Internal phase: *think of a hand on a clock, or an oven timer*



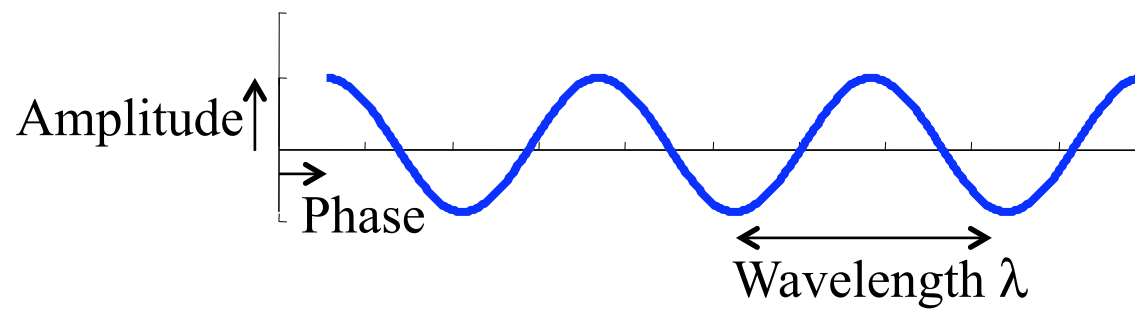
The wave function

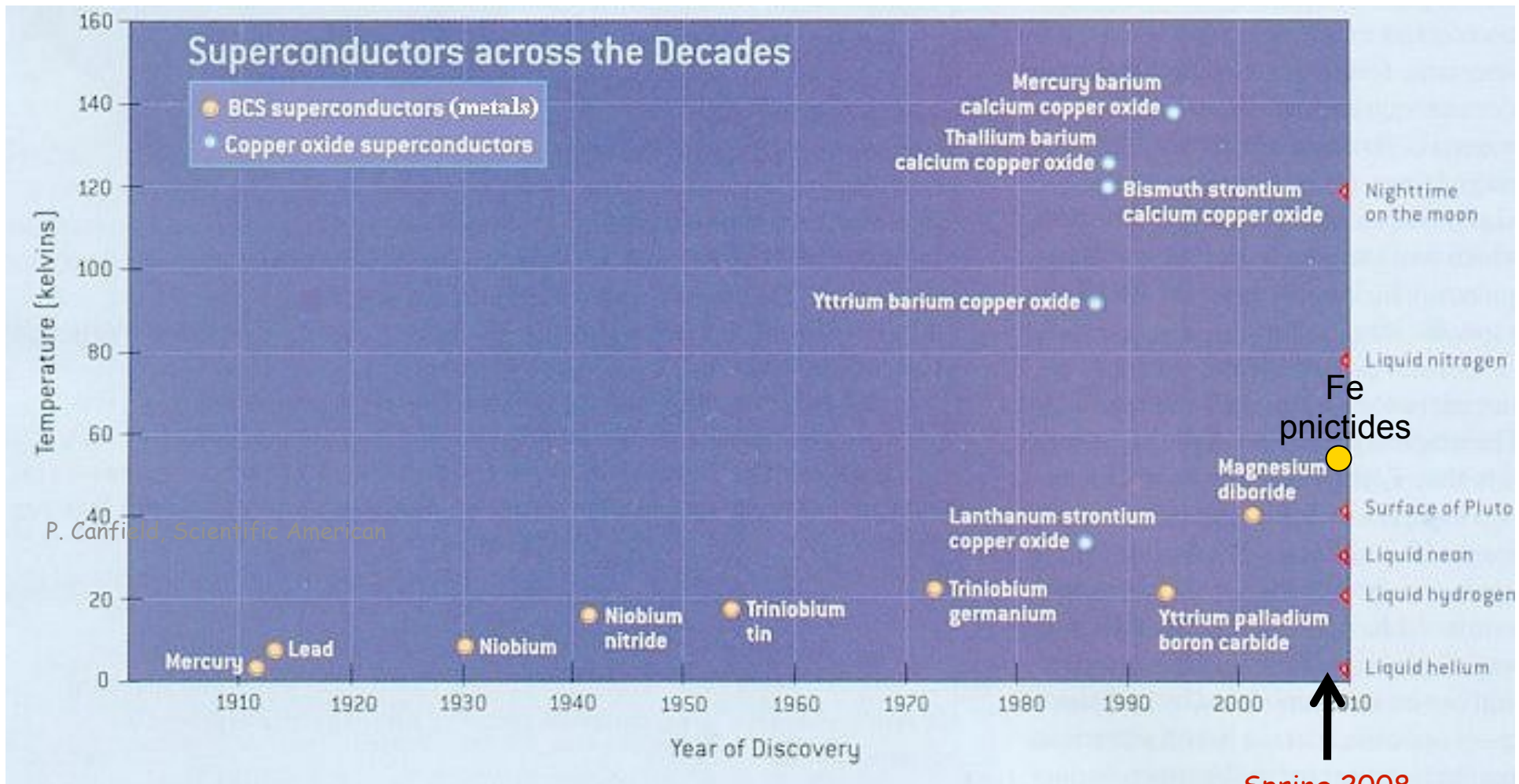
$$\Psi(r, t)$$

has an internal phase

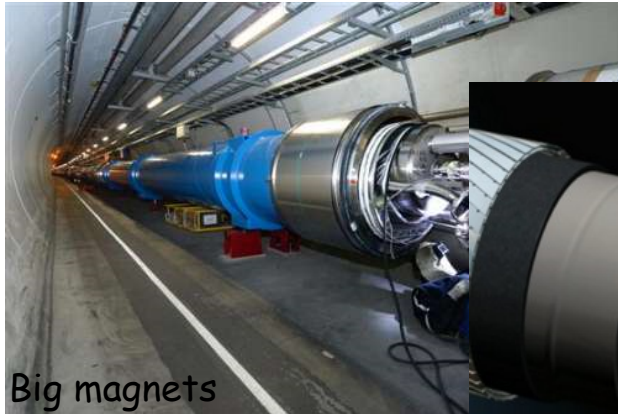
$$\Psi = \underbrace{\psi}_{\text{amplitude}} e^{i\phi} \quad \text{phase}$$

Mathematical example of a wave



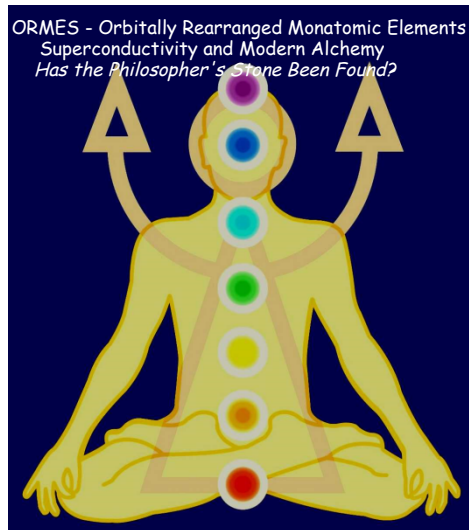


What is superconductivity good for ?

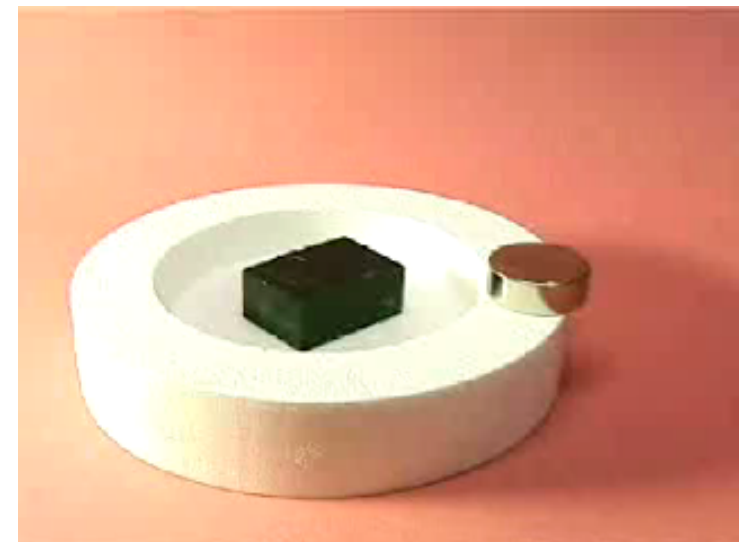


- Magnets in MRI systems
- Compact superquiet EM motors
- Fast information processing
- Directed energy devices
- ...

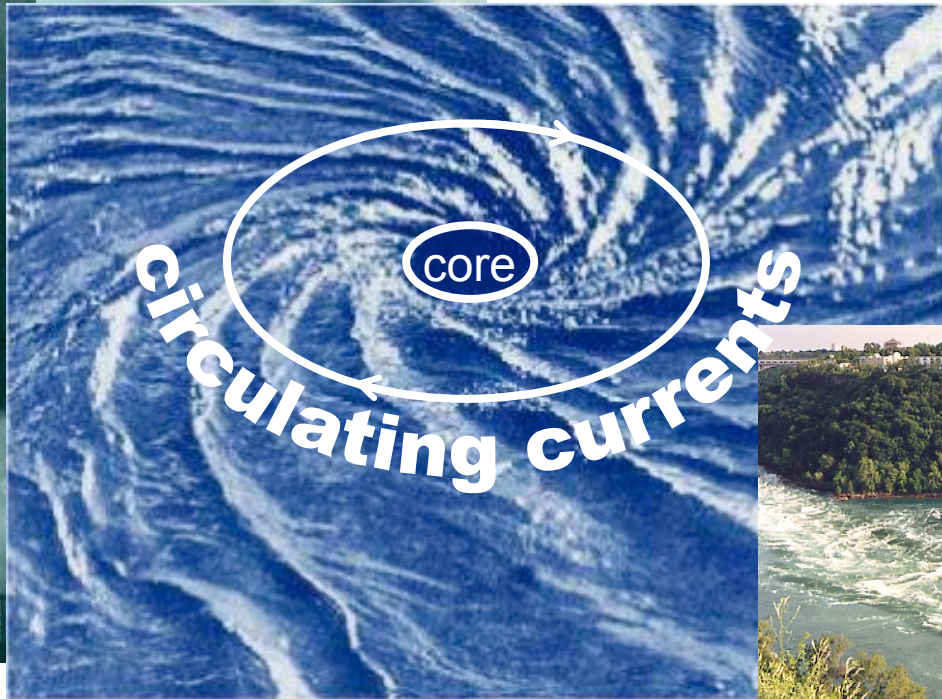
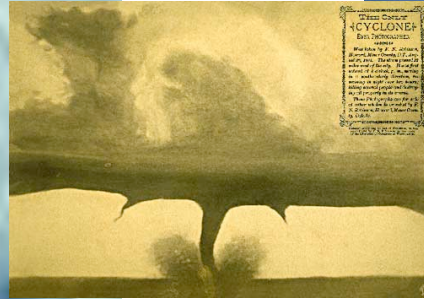
Pursuit of happiness



Physics Fairs



Vortices in Classical Systems



Vortices in Superconductors

Vortices in Normal and Quantum Fluids

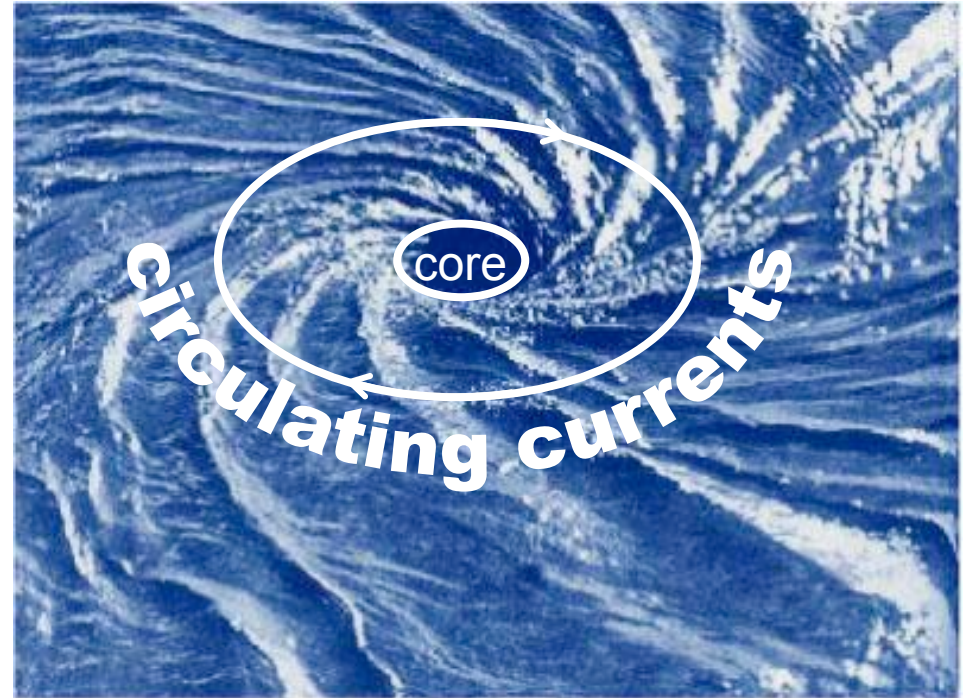
- Have cylindrical symmetry
- Consist of fluid swirling around a core, with the fluid's velocity getting faster and faster near the core

Vortices in Quantum Fluids

- Are persistent
- Are quantized

Vortices in Superconductors (which are charged quantum fluids)

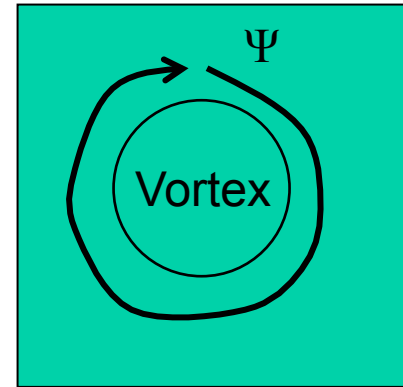
- Carry their own magnetic field
- Cause dissipation when they move
- Are the major limitation to the practical applications of superconductors



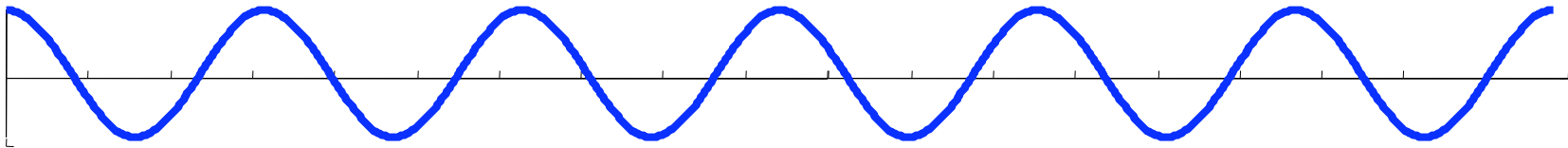
Each Vortex Carries *Quantized* Flux

The “pseudowavefunction” has a phase

$$\Psi = |\Psi| e^{i\phi}$$



For the superfluid swirling around the vortex core, the phase can only change by multiples of 2π

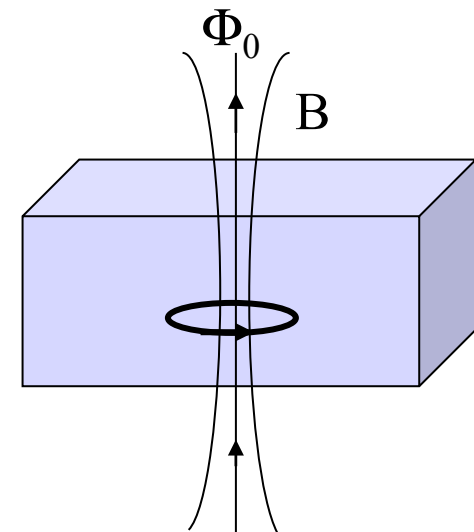


This leads to the quantization of magnetic flux:

$$\Phi = \int \vec{B} \cdot d\vec{a} = n\Phi_0$$

Superconducting flux quantum

$$\Phi_0 = hc/2e = 20.7 \text{ Gauss } \mu\text{m}^2$$

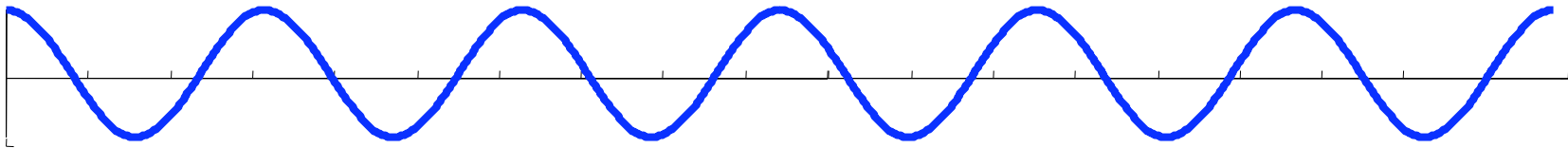
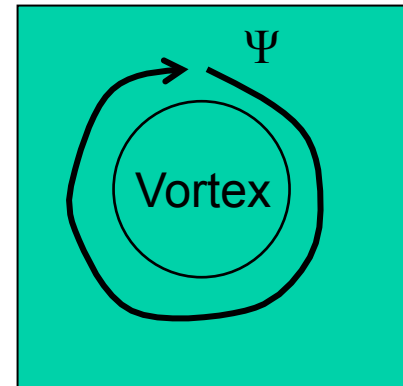


Why do we care about vortices in superconductors?

1) Fundamental interest

The vortex only exists because of the quantum “pseudowavefunction”

$$\Psi = |\Psi| e^{i\phi}$$



So studying the vortices often gives us clues about the nature of the underlying quantum state.